

**CHARACTERIZATION OF BROWN LAYERS ON FAÇADES OF  
ARCHAEOLOGICAL BUILDINGS IN SLOVENIA****Polonca Ropret<sup>1</sup>, Peter Bukovec<sup>2</sup>**<sup>1</sup>*Institute for the Protection of Cultural Heritage of Slovenia, Restoration Center,  
Poljanska 40, 1000 Ljubljana, Slovenia*<sup>2</sup>*Faculty of Chemistry and Chemical Technology, University of Ljubljana,  
Aškerčeva 5, 1000 Ljubljana, Slovenia***Darko Hanžel***J.Stefan Institute, Jamova 39, 1111 Ljubljana, Slovenia**Received 22-02-2000***Abstract**

Brown layers have occurred just below the façade surface of some archaeological buildings. They are a consequence of chemical changes, which can give rise to further destruction. Brown layers and undamaged plaster underneath the brown layers consist of CaCO<sub>3</sub>, SiO<sub>2</sub>, Ca<sub>2</sub>SiO<sub>4</sub>, CaMg(CO<sub>3</sub>)<sub>2</sub>, silicate minerals containing Ca, Al, Mg, K, Fe, metal hydroxides and sulphates, with some surface bound water. In both layers coordination of Fe<sup>3+</sup> ions is tetrahedral and octahedral. In brown layers there are more Fe<sup>3+</sup> ions at octahedral sites and there is an evidence of a change in chemical environment of ferric ions. The difference between brown layers and undamaged plaster underneath the brown layers is in goethite ( $\alpha$ -FeOOH) formation. It is microcrystalline and some of the Fe<sup>3+</sup> ions are substituted by a non magnetic ion, probably Al<sup>3+</sup>. Iron in oxidation state 2+ is present only in the brown layer of sample ZMV 7. One of the possible ways of its genesis is reduction of Fe<sup>3+</sup> due to SO<sub>2</sub> present in the air.

**Introduction**

A change of colour on archaeological objects often agitates restorers and architects because it is aesthetically unattractive and it can also give rise to further destruction of those objects. On some constructions of historical value in Slovenia, built or renewed around year 1910, brown layers just below the facade surface have occurred. Those layers are hard and thin, usually 1-2mm. There are parts where brown colour has entered even further into the interior of the plaster. In the Centre for Restoration of the Republic of Slovenia they have prepared colour studies<sup>1</sup> of samples taken from building facades. Those studies were done by microscopic observations. Since the borderline between brown layers and undamaged plaster beneath is not sharp, they assumed that brown

layers were not created by adding a pigment into the external layer of building facades, but are a consequence of chemical changes.

Some of the former investigations<sup>2,3,4</sup> of cements and iron containing material, have shown, that iron could be included into chemical changes of building plaster. D. Hanžel, D. Dimic and D. Lasič studied hydration behaviour of two Yugoslav cements by Mössbauer spectroscopy.<sup>2</sup> According to their investigation cement contains minerals with ferric ions at octahedral and tetrahedral sites in ferrites of brownmillerite-like structure having the generic formula  $2\text{CaO}\cdot(\text{Fe}_2\text{O}_3)_{1-x}\cdot(\text{Al}_2\text{O}_3)_x$ . Spectra of hydrated samples differ from those of nonhydrated samples. This indicates a change in chemical environment of ferric ions during the process of hydration. There is also an evidence that the ferrite phase might take part in the complex cementing action during ageing.

Analysis of thin black layers on building stone<sup>3,4</sup> showed that the layers mainly consist of various iron oxides and iron oxide hydroxides, sulphates, soot and silicate minerals, with smaller amounts of metal, rubber and asphalt particles. In low concentration are present also the organic constituents.

The aim of this investigation was to characterise brown layers formed on building facades. Various techniques including electronic microanalysis, thermogravimetric and differential thermal analysis, X-ray powder analysis, IR and Mössbauer spectroscopy were used to analyse these layers.

### **The samples**

Samples were taken from 4 archaeological buildings in Slovenia. Three of them are placed in Ljubljana (Zmaj bridge-sample ZMV 7, Barrier on river Ljubljanica-sample ZAP 10, Auersperger palace-sample AUE 37), and one in Domžale (Social home Domžale-sample DDD 36). From each of the four buildings one sample was analysed. In each case, a sample of corresponding undamaged plaster underneath the brown layer was collected for comparison.

Brown layers and undamaged plaster were separated by a diamond knife. The knife was used because the material is very hard, and to avoid contamination. For electronic microanalysis samples were prepared in a shape of a flat tile. For all other analyses samples were milled into powder and homogenised.

### **Instrumental analysis**

Electronic microanalyses were done on an electronic microscope JEOL JSM 35, equipped with energy disperse spectrometer Tracor. Thermo gravimetric and differential thermal analyses were taken on a Perkin-Elmer 7 Series Thermal Analysis system in air flow. The heating rate was 5 °C/min. X-ray powder patterns were taken on a Huber Guinier camera 620 (Cu K $\alpha$ ). The patterns were compared with standard patterns in the PDF<sup>5</sup> using  $\mu$ PDSM computer program.<sup>6</sup> Infrared spectra were taken on a Perkin-Elmer FT-IR 1720 X spectrometer in nujol matrix in CsI single crystal windows. Mössbauer spectra were measured by using a Wissel constant acceleration spectrometer. The spectra were analysed by a non-linear least squares fitting procedure. Metallic iron was used as a reference for isomer shift parameter as well as for calibration of the velocity scale. The  $\gamma$ -ray source was <sup>57</sup>Co in rhodium matrix.

### **Results**

Elemental composition was determined by electronic microanalysis. The surface of the samples is inhomogeneous. Therefore electronic microanalysis was used only as a qualitative analysis. As it is shown in Table 1 the main component in both layers is calcium. Other elements are present in smaller concentration. Sample AUE 37 – undamaged plaster is unsuitable for this investigation because it is in a form of fine dry particles and it was impossible to get it in a shape of a flat tile.

**TABLE 1.** The results of the electronic microanalysis given in weight percent considering oxides.

a.) brown layer

b.) undamaged plaster underneath the brown layer

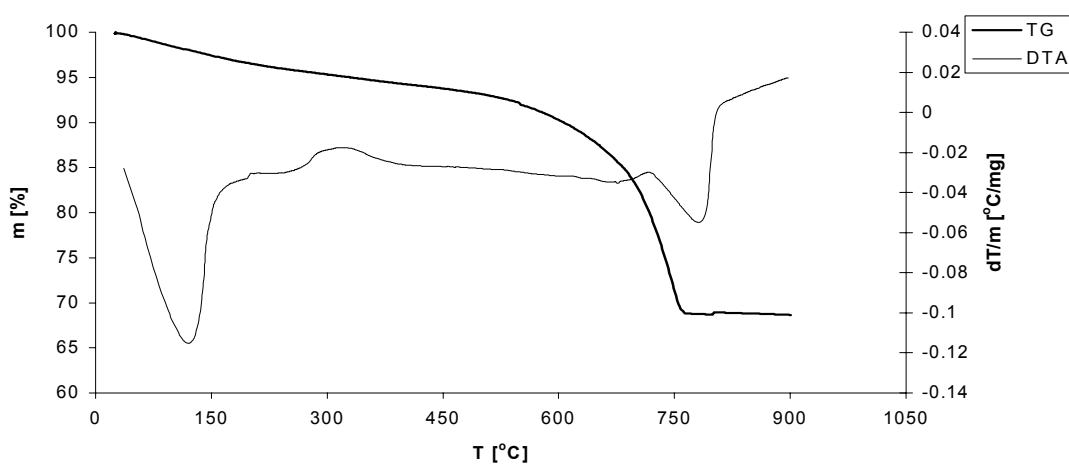
1., 2., 3., 4. – different locations of the same sample

Remark: sample AUE 37-b.) is unsuitable for this investigation

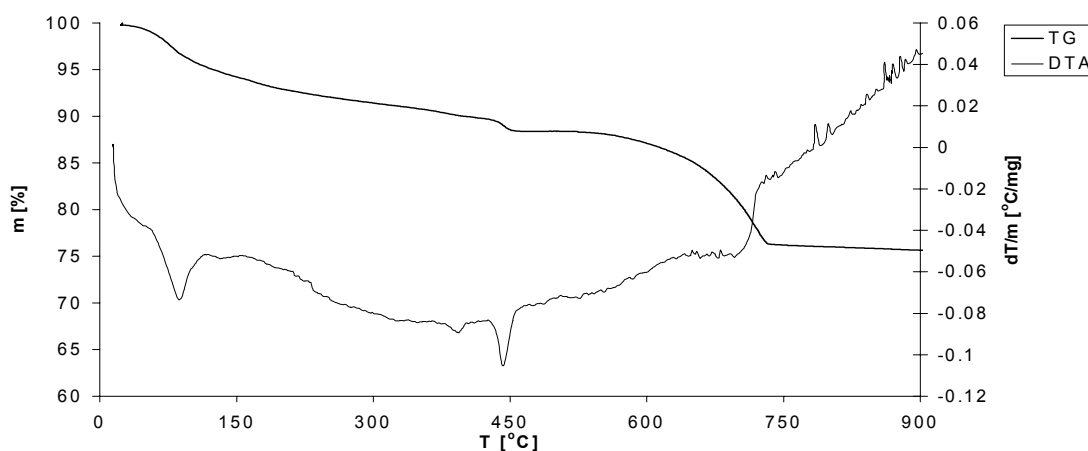
	Ca	Si	K	Mg	Al	S	Fe	Ti	P	Cl	
<b>DDD 36</b> a.) 1.	52.26	4.77	19.19	0.09	3.89	12.91	5.68	0.09			
	77.61	10.04	4.34	1.09	1.86	1.53	1.89	0.48			
	50.80	2.85	0.77	0.53	12.05			1.03			
	b.) 1.	67.64	19.47		0.79	4.76	0.29	5.76	0.21		
		87.39	6.58		0.46	1.93	1.37	0.87	0.16		
<b>ZMV 7</b> a.) 1.	5.03	86.47	1.94		2.05		3.30				
	80.80	8.09	1.98	1.54	2.48	1.03	1.65				
	80.25	6.77	1.45	2.66	1.54	1.23	2.87	0.65	1.28		
	73.99	2.25		17.29	0.38	0.85	1.65		0.86		
	b.) 1.	2.60	96.47								
		70.65	10.57		8.04	4.49	2.15	2.68			
		75.39	1.21		21.99						
		69.23	18.13	0.98	5.29	2.64	0.72	1.73			
	<b>ZAP 10</b> a.) 1.	80.93	10.82	0.63		1.04	2.92	1.24			0.59
		23.76	39.94	4.71		5.22	18.98	5.85	0.47		
44.42		9.98			2.87	39.34	0.31				
b.) 1.		96.52	1.50							0.64	
		41.57	40.62	1.28	0.95	5.68	2.10	5.77	0.98		
		93.53	2.63	0.70	0.24	0.39	0.85	0.17	0.10		
<b>AUE 37</b> a.) 1.		74.53	10.51	2.51	1.03	3.07	2.94	4.13			
	68.77	12.49	4.28	1.61	3.29	3.54	5.05				
	64.34	13.08	6.92	0.90	3.29	3.23	6.99				

Thermo gravimetric and differential thermal analyses gave similar results for all samples as well as for corresponding undamaged plaster underneath. In both cases TG curves show two significant mass losses, one up to 100 °C and another one more intense in the temperature range of 600-900 °C (Figure 1). At these temperatures DTA curves show two corresponding endothermic peaks. In TG curves a slow, continuous weight loss is noticeable in the temperature range of 100-500 °C. TG analysis of samples DDD 36-undamaged plaster (Figure 2) and ZMV 7-undamaged plaster shows another mass loss at temperature 400-450 °C.

**FIGURE 1.** TG and DTA analysis of the sample DDD 36 – brown layer



**FIGURE 2.** TG and DTA analysis of the sample DDD 36 – undamaged plaster underneath the brown



Infrared spectra were taken on a sample DDD 36 – brown layer and on a residual after TG analysis of the same sample. Characteristics:

1. DDD 36 – brown layer

- a.) Broad band in the area of  $3700 - 3000 \text{ cm}^{-1}$ .
- b.) Strong, broad band ranging from  $1500$  to  $1400 \text{ cm}^{-1}$ .
- c.) Broad band in the area of  $1200 - 800 \text{ cm}^{-1}$  with one sharp peak at  $873 \text{ cm}^{-1}$ .
- d.) Strong, sharp peak at  $713 \text{ cm}^{-1}$ .

2. The rest after TG analysis of sample DDD 36 – brown layer

- a.) Sharp band at  $3640 \text{ cm}^{-1}$ .
- b.) Broad band in the area of  $1200 - 800 \text{ cm}^{-1}$ .

X-ray powder analyses were done on all samples and on TG residuals. Powder patterns show many broadened or overlapping diffraction lines. It is therefore difficult to determine the exact composition of samples. Components which show the best matching of lines comparing to lines of components from PDF computer database are:  $\text{CaCO}_3$ ,  $\text{SiO}_2$ ,  $\text{Ca}_2\text{SiO}_4$  and  $\text{CaMg}(\text{CO}_3)_2$ . It was often easier to determine composition of the rest after TG analysis because the lines are sharper and less overlapping. Tables from 2 to 17 show the results of x-ray powder analysis.

Mössbauer spectra have been recorded on brown layers of samples DDD – 36 and ZMV – 7 and on the undamaged plaster underneath brown layers of these samples. Spectra of undamaged plaster of both samples consist of two overlapping doublet patterns (Figure 3). Similar patterns are present also in spectra of brown layers (Figure 4) except there is a change in quadrupole splitting parameter of approximately  $0.2 \text{ mm/s}$  and the ratio of relative intensities also becomes different. In the spectrum of sample ZMV – 7-brown layer (Figure 5) another quadrupole splitting doublet is present, with higher values for isomer shift and quadrupole splitting parameter but a very low intensity. Hyperfine parameters of Mössbauer spectra recorded at room temperature are given in Table 18 and 19.

**TABLE 2.** Diffraction lines (d) and relative intensities ( $I/I_0$ ) of the sample ZMV 7-brown layer compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE		CaCO <sub>3</sub> (Calcite) 5-0586*		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 33-0302*		CaMg(CO <sub>3</sub> ) <sub>2</sub> (Dolomite) 36-0426*		Ca <sub>2</sub> Al <sub>2</sub> Si <sub>2</sub> O <sub>15</sub> H <sub>2</sub> (Zoisite) 13-0562I	
d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$
3.324	90			3.343	214	3.322	89			3.068	19
3.037	100	3.035	199			3.021	8.9			2.874	42
2.889	80					2.895	4.4	2.888	222	2.790	19
2.778	30					2.775	8.9			2.720	6.4
2.698	30					2.671	18	2.670	9		
2.608	10					2.588	4.4	2.539	7		
2.536	10							2.193	42	2.536	6.4
2.494	40	2.495	28			2.302	4.4			2.516	3.2
2.284	60	2.285	36	2.282	26	2.242	4.4			2.280	6.4
2.236	10										
2.191	50										
2.127	10			2.128	19					2.101	13
2.111	10										
2.096	50	2.095	36			2.019	4.4	2.015	22	2.019	22
2.015	10					1.982	8.9			1.984	13
1.977	20			1.980	13						
1.925	20	1.927	10								
1.910	40	1.913	34			1.912	18				
1.876	40	1.875	34			1.878	35				
1.846	10					1.836	27				
1.816	40			1.817	36						
1.805	10			1.801	2			1.805	22		
1.788	40							1.787	29		
1.604	10	1.604	16	1.608	2					1.601	22
1.541	30			1.541	32					1.541	13
1.525	10	1.525	10								
1.381	10			1.382	15						
1.373	10			1.375	24			1.388	4		

**TABLE 3.** Diffraction lines (*d*) and relative intensities (*I*/*I*<sub>0</sub>) of the sample ZMV 7-undamaged plaster compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE ZMV 7- undamaged plaster	CaCO <sub>3</sub> (Calcite) 5-0586*		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 33-0302*		CaMg(CO <sub>3</sub> ) <sub>2</sub> (Dolomite) 36-0426*		Ca <sub>2</sub> Al <sub>3</sub> Si <sub>3</sub> O <sub>12</sub> (OH) (Clinozoisite) 13-0563D	
	<i>d</i> [Å]	<i>I</i> / <i>I</i> <sub>0</sub>	<i>d</i> [Å]	<i>I</i> / <i>I</i> <sub>0</sub>	<i>d</i> [Å]	<i>I</i> / <i>I</i> <sub>0</sub>	<i>d</i> [Å]	<i>I</i> / <i>I</i> <sub>0</sub>	<i>d</i> [Å]	<i>I</i> / <i>I</i> <sub>0</sub>
3.324	90	3.343	97	3.049	4.8	3.033	21	3.033	21	
3.037	40	3.035	66	2.877	11	2.888	193	2.885	60	
2.888	100			2.790	52	2.790	8	2.789	21	
2.777	30			2.718	16	2.670	6	2.669	18	
2.695	30			2.610	23	2.539	14	2.634	9.6	
2.637	10			2.545	4.8	2.404	37	2.587	18	
2.607	10			2.410	7	2.193	19	2.397	12	
2.540	10			2.301	2.2	2.015	4	2.300	2.4	
2.402	30	2.285	12	2.282	12	1.805	19	2.097	15	
2.282	10	2.095	12			1.787	25	2.014	4.8	
2.192	50	1.927	3.3			1.388	4			
2.105	10	1.913	11							
2.015	40	1.817	16							
1.929	10	1.525	3.3							
1.905	10	1.382	6.8							
1.817	30	1.375	11							
1.802	10									
1.785	30									
1.525	10									
1.381	10									
1.373	10									



**TABLE 4.** Diffraction lines ( $d$ ) and relative intensities ( $I/I_0$ ) of the rest after TG analysis of the sample ZMV 7-brown layer compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG ZMV 7- brown layer		CaO (Lime) 43-100ID		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 24-0037D		Ca <sub>54</sub> MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub> 13-0272		CaFe <sub>3</sub> O <sub>5</sub> 30-0256		CaSO <sub>4</sub> (Anhydrite) 6-0226D		Mg <sub>2</sub> SiO <sub>4</sub> 34-0556	
$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$
4.238	20			4.260	81	1.264	4.8					3.498	111	3.460	4.1
3.485	40			3.343	230	3.175	4.8	3.039	35	3.210	5			3.340	8.3
3.330	100					3.048	11	2.973	14			3.118	4	3.210	33
3.178	40							2.968	14						
3.109	10							2.776	71	2.810	27				
3.029	10							2.764	42						
2.997	10							2.748	28						
2.965	10							2.732	49						
2.884	10							2.717	94						
2.841	10									2.660	109			2.850	17
2.795	10									2.620	33				
2.769	70	2.777	17							2.490	27				
2.741	70									2.390	11				
2.689	60														
2.662	30														
2.625	40					2.610	74	2.606	50					2.620	41
2.598	40					2.582	18	2.608	28					2.510	17
2.505	10					2.503	12							2.480	17
2.471	10					2.493	26							2.440	83
2.449	40			2.458	28	2.452	2.9								
2.396	50														
2.320	10					2.305	17	2.323	7.1					2.270	4.1
2.276	50			2.282	28	2.282	35							2.230	17
2.229	20			2.237	14									2.200	33
2.200	40					2.189	61								
2.183	40					2.166	10	2.184	28					2.208	9
2.119	50			2.128	21	2.132	2.9	2.180	28					2.183	

TABLE 4 CONTINUE

The rest after TG ZMV 7- brown layer		CaO (Lime) 43-100ID		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 24-0037D		Ca <sub>8</sub> MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub>		CaFe <sub>3</sub> O <sub>5</sub>		CaSO <sub>4</sub> (Anhydrite) 6-0226D		Mg <sub>2</sub> SiO <sub>4</sub>	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.100	70					2.106	18			2.107	71	2.086	11		
2.043	20					2.047	20							2.020	83
2.020	20			1.980	14	2.026	5.7			1.973	27	1.993	7		
1.978	10					1.984	22							1.940	4.1
1.968	10									1.977	7.1	1.938	4		
1.938	30									1.983	7.1	1.938	4	1.910	8.3
1.923	30					1.913	5.7			1.938	7.1	1.938	4	1.820	1.7
1.901	10					1.894	28			1.923	7.1				
1.834	10					1.847	5.7								
1.814	70			1.817	39	1.821	4.8			1.824	7.1				
1.798	20			1.801	2	1.809	17			1.823	7.1				
1.790	20					1.791	11								
1.767	10					1.767	10			1.766	28			1.760	12
1.751	10					1.762	6.7			1.761	28	1.749	13	1.740	8.3
1.696	30	1.701	23			1.707	14								
1.687	30									1.699	21				
1.644	10					1.635	7.6			1.650	3	1.648	15	1.640	8.3
1.628	20					1.628	8.6								
1.573	10									1.581	2			1.570	1.7
1.565	10											1.564	7	1.560	8.3
1.541	30			1.541	35					1.546	20			1.541	17
1.499	50									1.508	10				
1.490	10											1.490	7		
1.486	50														
1.463	10									1.469	22				
1.455	10			1.453	7										
1.449	10	1.450	6												
1.381	10	1.389	6	1.382	10										
1.373	10			1.375	25										
1.370	40			1.372	21										
1.255	20			1.255	9							1.365	2		

TABLE 4 CONTINUE

The rest after TG ZMV 7- brown layer		CaO (Lime) 43-1001D		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 24-0037D		Ca <sub>54</sub> MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub>		CaFe <sub>3</sub> O <sub>5</sub>		CaSO <sub>4</sub> (Anhydrite) 6-0226D		Mg <sub>2</sub> SiO <sub>4</sub>	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.238	10											1.237	2		
1.227	30			1.228	5							1.216	4		
1.214	10											1.199	2		
1.204	10	1.203	2.3	1.200	12							1.178	1		
1.183	10			1.184	9							1.166	4		
1.179	10			1.180	9										
1.162	10														

TABLE 5. Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of the rest after TG analysis of the sample ZMV 7-undamaged plaster compared to lines of the most probable components from μPDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG ZMV 7- undamaged plaster		CaO (Lime) 43-1001D		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		Mg <sub>3</sub> Al <sub>2</sub> Si <sub>3</sub> O <sub>12</sub>		MgO (Periclase) 43-1022C		Ca <sub>2</sub> (Al,Fe) <sub>2</sub> O <sub>5</sub> (Brownillerite) 30-0226*		Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub>	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
4.905	30					4.900	12							4.409	3.4
4.406	10							4.460	4.3					4.215	2.4
4.259	20			4.260	56										
3.493	10													3.352	2.4
3.346	90			3.343	159	3.377	15								
3.269	20					3.241	17								
3.137	10					3.175	15								
3.117	10														
3.074	10					3.046	17							3.128	0.5
2.886	20					2.876	43							3.055	3.4
2.827	70					2.814	25								
2.782	70	2.777	35			2.795	124					2.784	14		
2.753	100					2.780	112								
2.731	20					2.744	118							2.714	31
2.696	90					2.716	50							2.693	48
2.668	40														

TABLE 5 CONTINUE

The rest after TG ZMV 7- undamaged plaster		CaO (Lime) 43-1001D		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		Mg <sub>3</sub> Al <sub>2</sub> Si <sub>3</sub> O <sub>12</sub> 30-0788O		MgO (Periclase) 43-1022C		Ca <sub>2</sub> (Al,Fe) <sub>2</sub> O <sub>5</sub> (Browmillerite) 30-0226*		Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub> 32-0150*	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.640	60					2.608	81	2.592	81			2.644	54		
2.608	60			2.458	19	2.453	25	2.440	13	2.432	7.9	2.472	1.1		
2.455	30	2.405	95			2.407	25	2.380	34						
2.399	70			2.237	10									2.206	9.6
2.236	10													2.186	1.9
2.201	50					2.196	15								
2.192	50					2.130	15								
2.126	30			2.128	14	2.099	110	2.097	86	2.106	73	2.210	4.3		
2.106	60					2.048	25					2.051	19	2.048	1.0
2.049	20					2.041	15							2.032	1.0
2.040	20			1.980	10	1.983	43								
1.998	80													1.963	1.4
1.980	10														
1.966	10														
1.938	50														
1.928	20					1.913	12	1.930	4.3			1.928	19	1.919	17
1.908	20			1.817	27	1.821	7							1.906	1.4
1.819	40			1.801	2	1.808	15					1.815	24	1.828	0.5
1.803	40					1.791	15					1.795	1.6	1.797	0.5
1.791	40					1.768	7							1.768	0.5
1.768	30							1.752	34						
1.757	30	1.701	47			1.707	17							1.699	0.5
1.699	50														
1.691	30			1.672	11	1.661	5								
1.672	10			1.659	5	1.634	17	1.623	77					1.625	1.4
1.655	10					1.575	15					1.578	7.6	1.564	17
1.629	50					1.552	12	1.546	4.3			1.564	2.2	1.551	9.6
1.575	10			1.541	24							1.538	7.6		
1.555	10											1.501	4.3		
1.541	30														
1.504	10														

TABLE 5 CONTINUE

The rest after TG ZMV 7- undamaged plaster		CaO (Lime) 43-1001D		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		Mg <sub>3</sub> Al <sub>2</sub> Si <sub>3</sub> O <sub>12</sub> 30-0788O		MgO (Periclase) 43-1022C		Ca <sub>2</sub> (Al,Fe) <sub>2</sub> O <sub>5</sub> (Brownmillerite) 30-0226*		Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub> 32-0150*	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.489	50					1.485	10			1.490	37			1.493	1.4
1.411	10			1.418	2	1.417	12	1.410	30					1.414	0.5
1.392	10	1.389	12			1.397	7	1.392	4.3			1.390	3.2		
1.383	10			1.382	11										
1.375	10			1.375	18			1.373	39						
1.349	10														
1.272	10							1.269	4.3	1.270	4.3				
1.262	40			1.256	6			1.258	4.3	1.216	11				
1.216	10														
1.200	30	1.203	4.8	1.200	8										
1.152	40			1.153	3									1.347	5.8

TABLE 6. Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of sample DDD 36-brown layer compared to lines of the most probable components from μPDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE	CaCO <sub>3</sub> (Calcite) 5-0586*		CaO (Lime) 28-0775	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
DDD 36- brown layer	4.1511	30		
	3.8619	10	3.86	21
	3.3408	30		
	3.0260	100	3.035	171
	2.4837	30	2.495	24
	2.2774	50	2.285	31
	2.0894	40	2.095	31
	1.9061	40	1.913	29
	1.8701	40	1.875	29
			3.33	69
			3.00	69
			2.28	41
			2.09	28
			1.91	41
			1.88	41

Remark: The presence of CaO is questionable.

**TABLE 7.** Diffraction lines (d) and relative intensities ( $I/I_0$ ) of sample DDD 36-undamaged plaster compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE	CaCO <sub>3</sub> (Calcite) 5-0586*		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0951D		Ca <sub>2</sub> (Al,Fe) <sub>2</sub> O <sub>5</sub> (Brownmillerite) 30-0226*	
	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$
DDD 36- undamaged plaster	2.9992	100	3.035	178		
	2.7915	30			2.784	5.5
	2.7025	30			2.731	11
	2.6532	10			2.673	7.7
	2.6186	30			2.608	17
	2.5023	40	2.495	25		
	2.2904	50	2.285	32		
	2.1960	20			2.304	1.6
	2.1005	40	2.095	32	2.210	1.8
	1.9141	40	1.913	30	1.9283	7.7
	1.8795	40	1.875	30	1.8813	0.7
1.8194	10			1.8149	9.9	

**TABLE 8.** Diffraction lines (d) and relative intensities ( $I/I_0$ ) of the rest after TG analysis of the sample DDD 36-brown layer compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG DDD 36- brown layer	CaO (Lime) 43-1001D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 33-0302*		Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub>		MgO (Periclase) 43-1022C		Ca <sub>1,7</sub> Mg <sub>0,3</sub> SiO <sub>4</sub> (Bredigite) 35-0260D	
	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$
4.9252	10		4.892	1.4*					3.38	7.1
3.525	10				3.3312	1.3*			2.959	2.8
2.9307	10				2.938	0.7*			2.837	9.2
2.8488	10									
2.8259	10		2.814	10	2.8342	3.3			2.783	6.3
2.7790	50	2.777	2.790	46	2.7871	5.4*			2.736	71
2.7503	40		2.745	39						

TABLE 8 CONTINUE

The rest after TG DDD 36- brown layer		CaO (Lime) 43-1001D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 33-0302*		Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub> 38-1429*		MgO (Periclase) 43-1022C		Ca <sub>1,7</sub> Mg <sub>0,3</sub> SiO <sub>4</sub> (Bredigite) 35-0260D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.6994	40					2.6987	67			2.672	68
2.6384	30			2.610	20					2.599	6.3
2.6086	30			2.403	8.6	2.4135	3.3*			2.424	4.9
2.4032	100	2.405	171	2.281	10	2.2752	2.0*				
2.2837	10					2.2506	0.7*			2.233	19
2.2409	10									2.215	14
2.2256	10										
2.1906	30			2.189	24					2.176	7.8
2.1675	10			2.165	6.2					2.113	6.3
2.1030	20			2.103	0.5*			2.106	29	2.081	4.9
2.0815	20			2.083	2.9*						
2.0534	10			2.050	6.7					2.028	6.3
2.0238	10			2.027	7.1					2.004	9.2
1.9955	10										
1.9876	10			1.987	9.5					1.954	4.9
1.9431	20									1.924	20
1.9264	20									1.899	11
1.9062	10			1.9115	2.9	1.9079	20				
1.7000	80	1.7008	84	1.6964	2.4					1.622	1.4
1.6320	10			1.6282	5.7	1.6270	0.7*			1.576	4.9
1.5748	10			1.5738	2.4	1.5744	0.7*			1.548	4.9
1.5554	20					1.5578	16				
1.4875	10					1.4895	0.7*	1.4895	15		
1.4496	40										
1.4106	10	1.4504	22								
1.4054	10					1.4110	0.7*				
1.3984	10			1.4050	0.7*					1.399	4.9
1.3883	30	1.3887	22			1.3876	0.7*			1.386	1.4

TABLE 8 CONTINUE

The rest after TG DDD 36- brown layer		CaO (Lime) 43-1001D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 33-0302*		Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub> 38-1429*		MgO (Periclase) 43-1022C		Ca <sub>1.7</sub> Mg <sub>0.3</sub> SiO <sub>4</sub> (Bredigite) 35-0260D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.3755	10					1.3760	0.7*				
1.3489	10					1.3490	6				
1.2615	20										
1.2158	10					1.2142	0.7*	1.2162	4.4		
1.2022	20	1.2026	9								
1.1518	10										
1.1031	10	1.1036	9								
1.0753	30	1.0757	22								

TABLE 9. Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of the rest after TG analysis of the sample DDD 36-undamaged plaster compared to lines of the most probable components from μPDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG DDD 36- undamaged plaster		Ca <sub>2</sub> (Al,Fe) <sub>2</sub> O <sub>5</sub> (Brownmillerite) 30-0226*		Cs <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		SiO <sub>2</sub> (Quartz) 5-0490D		Fe <sub>2</sub> O <sub>3</sub> (Hematite) 13-0534D		CaSO <sub>4</sub> 43-0606I		Ca <sub>1.82</sub> Al <sub>3.64</sub> Si <sub>0.38</sub> O <sub>8</sub> 26-0559	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
4.9054	30			4.90	15								
4.2933	20					4.26	31			4.360	4.4	4.09	13
4.1381	20												
4.0073	20			3.827	12								
3.7979	20			3.783	12							3.84	28
3.7529	10												
3.6817	10	3.654	29					3.66	13				
3.4804	40									3.485	70		
3.3637	30	3.406	7	3.377	18								
3.3317	30					3.343	89						
3.2203	10			3.241	21								
3.0267	10			3.046	21								
2.9993	60									3.018	79	2.977	51



TABLE 9 CONTINUE

The rest after TG DDD 36- undamaged plaster		$\text{Ca}_2(\text{Al,Fe})_2\text{O}_5$ (Brownmillerite) 30-0226*		$\text{Cs}_2\text{SiO}_4$ (Larnite) 9-0351D		$\text{SiO}_2$ (Quartz) 5-0490D		$\text{Fe}_2\text{O}_3$ (Hematite) 13-0534D		$\text{CaSO}_4$ 43-0606I		$\text{Ca}_{1.82}\text{Al}_{3.64}\text{Si}_{0.38}\text{O}_8$ 26-0559	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.8836	30			2.814	30					2.797	44	2.847	4.1
2.8389	40	2.784	46	2.780	136					2.723	17		
2.7716	90			2.744	144			2.69	54				
2.7378	100			2.716	61								
2.7187	90												
2.7088	10												
2.6835	60	2.673	64					2.51	27				
2.6322	60	2.644	184										
2.5982	80	2.576	31	2.608	98								
2.5314	10			2.543	21								
2.5050	10											2.509	30
2.4447	20			2.451	30	2.458	11						
2.4281	10	2.434	7	2.433	12								
2.3960	80			2.407	30							2.400	3.0
2.3187	40			2.322	9					2.339	8.7	2.307	9.1
2.2787	70			2.282	53			2.285	1.1	2.282	4.4		
2.2353	10							2.201	16				
2.1962	10	2.210	15	2.196	18							2.205	2.0
2.1815	40			2.188	98					2.182	8.7	2.173	3.0
2.1607	10	2.155	16	2.163	30								
2.1490	10									2.146	17		
2.1244	10			2.130	18							2.138	2.0
2.0985	20			2.094	12					2.104	4.4	2.100	2.0
2.0785	10			2.083	12								
2.0446	60	2.051	64	2.048	30			2.070	1.1			2.042	4.1
2.0335	10			2.041	18								
2.0195	40			2.019	30								
1.9915	10									2.013	17		
1.9790	40			1.983	53					1.986	4.4		

TABLE 9 CONTINUE

The rest after TG DDD 36- undamaged plaster	$\text{Ca}_2(\text{Al,Fe})_2\text{O}_5$ (Brownmillerite) 30-0226*		$\text{Cs}_2\text{SiO}_4$ (Larnite) 9-0351D		$\text{SiO}_2$ (Quartz) 5-0490D		$\text{Fe}_2\text{O}_3$ (Hematite) 13-0534D		$\text{CaSO}_4$ 43-0606I		$\text{Ca}_{1.82}\text{Al}_{3.64}\text{Si}_{10.38}\text{O}_8$ 26-0559	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.9424	30											
1.9320	30	1.9283	64	1.913	15			1.917	17	1.919	13	
1.9161	40			1.897	9							
1.9042	50											
1.8653	10	1.8632	16	1.845	9			1.838	22	1.846	2.0	
1.8367	20			1.821	9	1.817	15					
1.8126	10	1.8149	82	1.808	18	1.801	0.9					
1.7995	40			1.791	18							
1.7882	20	1.7952	5	1.768	9							
1.7661	10											
1.7504	40	1.7327	13	1.727	9			1.743	26			
1.7359	10			1.718	9							
1.7266	10			1.707	21			1.697	26	1.699	5.1	
1.6977	30							1.680	8.7			
1.6861	40			1.634	21			1.634	2.2			
1.6327	30			1.627	12			1.596	8.6			
1.6279	30			1.603	9	1.608	0.9					
1.6007	10			1.587	9					1.584	2.0	
1.5841	10	1.5784	26	1.575	15							
1.5713	20	1.5638	7	1.552	15							
1.5528	30											
1.5378	20	1.5380	26	1.523	12	1.541	13	1.546	4.4			
1.5255	10	1.5202	13					1.526	4.4	1.529	7.1	
1.5116	10											
1.4999	10	1.5013	15					1.509	26			
1.4915	50											
1.4824	40			1.485	12			1.484	19			
1.4699	20											

TABLE 9 CONTINUE

The rest after TG DDD 36- undamaged plaster		$\text{Ca}_2(\text{Al,Fe})_2\text{O}_5$ (Brownmillerite) 30-0226*		$\text{Cs}_2\text{SiO}_4$ (Larnite) 9-0351D		$\text{SiO}_2$ (Quartz) 5-0490D		$\text{Fe}_2\text{O}_3$ (Hematite) 13-0534D		$\text{CaSO}_4$		$\text{Ca}_{1.82}\text{Al}_{3.64}\text{Si}_{0.38}\text{O}_8$	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.4476	10	1.4524	7	1.445	12	1.453	2.7	1.452	19	1.437	4.4	1.449	13
1.4361	10					1.375	9.7						
1.3744	10					1.372	8.0						
1.3708	10	1.3669	5			1.256	3.5	1.258	4.3			1.369	2.0
1.3661	10	1.3588	9			1.1802	3.5						
1.2561	10					1.1408	0.9*	1.141	6.5				
1.2518	20												
1.1788	10												
1.1384	10												

TABLE 10. Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of the sample ZAP 10-brown layer compared to lines of the most probable components from μPDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE ZAP 10- brown layer		$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ (Gypsum) 21-0816*		$\text{CaCO}_3$ (Calcite) 24-0027D		$\text{Mg}_3\text{Ca}(\text{CO}_3)_4$ (Huntite) 14-409I	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
7.6179	90	7.61	108				
4.2806	100	4.28	216				
3.8506	10			3.852	43		
3.7996	20	3.80	19				
3.0668	80	3.07	72				
3.0312	80			3.030	148		
2.8747	70	2.871	240			2.888	44
2.8230	40			2.834	3*	2.833	222
2.7845	30	2.788	48				
2.7276	10					2.744	4
2.6808	70	2.684	120				

TABLE 10 CONTINUE

SAMPLE		$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ (Gypsum)		$\text{CaCO}_3$ (Calcite)		$\text{Mg}_3\text{Ca}(\text{CO}_3)_4$ (Humtite)	
ZAP 10- brown layer		21-0816*		24-0027D		14-409I	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.6394	10					2.604	27
2.5913	10	2.595	5				
2.5360	10						
2.4934	50	2.496	48	2.495	10*	2.432	22
2.4529	20	2.454	14			2.284	13
2.2817	50			2.284	27		
2.2169	50	2.220	14				
2.0882	60	2.087	34	2.094	40		
2.0734	50	2.073	48				
1.9946	30	1.993	5			1.991	22
1.9265	10			1.9261	6		
1.9101	20			1.9071	25		
1.8990	20	1.900	10			1.896	13
1.8757	40	1.880	14	1.8726	50		
1.8126	20	1.812	10			1.821	4
1.7982	20	1.798	14			1.796	4
1.7616	20					1.765	44
1.6662	10	1.664	10			1.656	2*
1.6444	10	1.646	5				
1.6212	30	1.622	10	1.6259	3*	1.611	2*
1.4391	20	1.440	10	1.4405	7	1.445	4
1.4196	10	1.418	5	1.4168	4	1.418	4
1.2616	10						
1.1526	10			1.1536	4		

TABLE 11. Diffraction lines (d) and relative intensities ( $I/I_0$ ) of the sample ZAP 10-undamaged plaster compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE		CaCO <sub>3</sub> (Calcite) 5-0586*		CaSO <sub>4</sub> ·2H <sub>2</sub> O (Gypsum) 36-0432D	
ZAP 10- undamaged plaster	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$
7.6179	30			7.59	112
4.2806	30			4.279	146
3.8506	40	3.86	39	3.795	19
3.3463	30				
3.0668	10			3.061	80
3.0312	100	3.035	321		
2.8855	30			2.875	61
2.7845	10			2.789	12
2.6812	10			2.683	45
2.4914	50	2.495	45	2.499	12
2.2817	50	2.285	58		
2.2143	10			2.215	13
2.1924	20				
2.0917	60	2.095	58	2.087	16
1.9250	30	1.927	16		
1.9094	60	1.913	55	1.8973	13
1.8731	60	1.875	55	1.8781	12
1.6244	30	1.626	13	1.6199	7
1.6031	40	1.604	26		
1.5239	40	1.525	16		
1.4391	40	1.440	16	1.4396	3*
1.4272	10	1.422	10		
1.4206	30				
1.1526	30	1.1538	10		

**TABLE 12.** Diffraction lines ( $d$ ) and relative intensities ( $I/I_0$ ) of the rest after TG analysis of the sample ZAP 10-brown layer compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG		CaSO <sub>4</sub> (Anhydrite) 6-0226D		CaO (Lime) 37-1497*		SiO <sub>2</sub> (Quartz) 33-1161*		CaAl <sub>12</sub> O <sub>9</sub>		CuO		Ca <sub>5</sub> MgSi <sub>3</sub> O <sub>12</sub>	
$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$	$d$ [Å]	$I/I_0$
4.9167	10					4.257	24	4.019	2.2*				
4.2650	20											3.891	1.8
4.0263	10												
3.8780	40	3.87	28										
3.7773	10												
3.4860	100	3.498	471			3.342	108					3.489	5.9
3.3495	30											3.377	7.1
3.2799	10											3.320	7.7
3.2172	10											3.227	3.5
3.1218	30	3.118	19										
3.0040	10							2.907	4.4			2.906	5.3
2.8997	10											2.871	2.4
2.8493	90	2.849	165									2.837	11
2.8251	10												
2.7973	30	2.797	19										
2.7741	50			2.7774	54			2.781	24	2.7752	7.6	2.758	3.5
2.7252	10							2.735	6.6			2.727	47
2.6980	20							2.695	1.3*	2.7047	6.0*	2.687	24
2.6367	20											2.655	15
2.6180	10							2.622	44			2.602	3.0
2.5822	10											2.581	5.9
2.5449	10									2.5444	28	2.558	4.7
2.5145	10											2.525	3.5
2.4710	50	2.473	38					2.479	44			2.454	1.8
2.4294	10											2.426	5.9
2.4040	80			2.4059	149			2.408	2.2			2.391	7.7
2.3468	10							2.352	8.7	2.3380	32	2.345	3.5

TABLE 12 CONTINUE

The rest after TG ZAP 10- brown layer		CaSO <sub>4</sub> (Anhydrite) 6-0226D		CaO (Lime) 37-1497*		SiO <sub>2</sub> (Quartz) 33-1161*		CaAl <sub>12</sub> O <sub>9</sub> 25-0122D		CuO 44-0706O		Ca <sub>5</sub> MgSi <sub>3</sub> O <sub>12</sub> 34-1350O	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.3270	90	2.328	94			2.282	9	2.287	13			2.322	8.3
2.2841	10					2.237	4					2.269	18
2.2646	10							2.211	4.4*			2.236	21
2.2379	10							2.188	4.4*			2.216	0.6
2.2087	90	2.208	94					2.110	26			2.169	9.5
2.1814	60	2.183	38					2.010	15			2.112	4.1
2.1654	30											2.086	9.5
2.1129	10											2.062	7.7
2.0844	70	2.086	47							2.0474	6.4	2.008	12
2.0472	10												
2.0162	10												
1.9938	60	1.993	28									1.928	24
1.9375	40	1.938	19									1.914	11
1.9230	10							1.908	1.3*			1.903	12
1.9034	10												
1.8845	10												
1.8687	90	1.869	75							1.8722	13*		
1.8523	40	1.852	19										
1.8411	10											1.841	11
1.8305	10												
1.8223	10					1.8179	15						
1.7986	10					1.8021	1*						
1.7798	10												
1.7702	20												
1.7491	90	1.749	56										
1.7346	10												
1.7157	10												
1.7004	70			1.7009	80			1.720	1.3				

TABLE 12 CONTINUE

The rest after TG ZAP 10- brown layer		CaSO <sub>4</sub> (Anhydrite) 6-0226D		CaO (Lime) 37-1497*		SiO <sub>2</sub> (Quartz) 33-1161*		CaAl <sub>12</sub> O <sub>9</sub> 25-0122D		CuO 44-0706O		Ca <sub>5</sub> MgSi <sub>3</sub> O <sub>12</sub> 34-1350O	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.6763	10					1.6719	4	1.681	1.3*				
1.6482	80	1.648	66			1.6082	1*	1.605	4.4				
1.6091	10												
1.5941	30	1.594	19					1.563	2.2*				
1.5649	30	1.564	28			1.5418	10	1.540	6.6				
1.5419	20							1.534	28				
1.5363	10							1.525	1.3*				
1.5251	50	1.525	19*					1.515	8.7	1.5102	11		
1.5147	10	1.515	9					1.487	1.3*				
1.4904	50	1.490	28					1.454	2.2*				
1.4508	40			1.4505	24	1.4536	1*			1.4164	10		
1.4432	10												
1.4386	10												
1.4249	40	1.424	19			1.4189	1*						
1.4187	20	1.418	9										
1.4119	10												
1.3973	40	1.398	19										
1.3886	40			1.3888	24			1.388	31				
1.3773	10					1.3752	8	1.380	4.4	1.3801	11		
1.3714	10					1.3718	9						
1.3651	10	1.365	9					1.347	1.3*				
1.3482	10												
1.3200	60	1.319	19*										
1.2977	10	1.296	9										
1.2769	60	1.277	28										
1.2298	10					1.2285	1*						
1.2227	10												
1.2188	10												



TABLE 12 CONTINUE

The rest after TG ZAP 10- brown layer		CaSO <sub>4</sub> (Anhydrite) 6-0226D		CaO (Lime) 37-1497*		SiO <sub>2</sub> (Quartz) 33-1161*		CaAl <sub>12</sub> O <sub>9</sub>		CuO		Ca <sub>2</sub> MgSi <sub>3</sub> O <sub>12</sub>	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.2165	30	1.216	19										
1.2082	10			1.2026	9*	1.1999	2			1.1968	7.3		
1.2022	30	1.1993	9			1.1843	3						
1.1992	10												
1.1904	10												
1.1843	10												
1.1731	10												
1.1656	40	1.1663	19										
1.1603	10												
1.1519	10												
1.1491	20	1.1483	9										
1.1434	10												
1.1417	10												
1.1071	40	1.1062	28										
1.1041	40	1.1044	19	1.1037	9*								
1.0956	10												
1.0927	10												
1.0888	10												
1.0838	10												
1.0789	30	1.0785	9*							1.0922	8.0		
1.0760	40												
1.0431	30			1.0758	24								
1.0405	30												
1.0313	10					1.0438	1*			1.0742	7.3		

**TABLE 13.** Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of the rest after TG analysis of the sample ZAP 10-undamaged plaster compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG ZAP 10- undamaged plaster		CaSO <sub>4</sub> (Anhydrite) 6-0226D		CaO (Lime) 37-1497*		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>5</sub> MgSi <sub>3</sub> O <sub>12</sub> 34-1350O		Ca(OH) <sub>2</sub> (Portlandite) 4-0733D		Ca <sub>8-11</sub> MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub> 13-0272	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
4.9167	10									4.9	50		
4.2691	10					4.26	13						
3.4995	50	3.498	82			3.343	38	3.500	2.2*				
3.3489	30												
3.2699	10												
3.1207	30	3.118	3.3*							3.112	15		
3.0947	30												
3.0325	10											3.034	11
3.0095	10												
2.9766	10												
2.8497	30	2.849	29					2.837	4.2*			2.973	4.6
2.8097	20	2.797	3.3*										
2.7762	70			2.7774	52								
2.7435	20												
2.7123	10												
2.6949	10												
2.6722	20												
2.6386	60												
2.6165	60												
2.4736	10	2.473	6.5										
2.4033	100			2.4059	144								
2.3296	20	2.328	16										
2.2956	10												
2.2668	10												
2.2397	10					2.237	2.3						
2.2079	20	2.208	16										
2.1880	40	2.183	6.5										
										2.628	67		
												2.608	16
												2.323	2.3
												2.184	9.2

TABLE 13 CONTINUE

The rest after TG ZAP 10- undamaged plaster		CaSO <sub>4</sub> (Anhydrite) 6-0226D		CaO (Lime) 37-1497*		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>5</sub> MgSi <sub>3</sub> O <sub>12</sub> 34-13500		Ca(OH) <sub>2</sub> (Portlandite) 4-0733D		Ca <sub>5</sub> [MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub> 13-0272	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.0487	10												
1.9471	40												
1.9381	40	1.938	3.3*									1.938	2.3
1.9199	40												
1.8879	10												
1.8703	20	1.869	13										
1.8016	50												
1.7928	50												
1.7488	10	1.749	9.8										
1.7282	10												
1.7003	80			1.7009	78								
1.6861	10												
1.6461	10	1.648	11										
1.5364	10												
1.4906	20	1.490	4.9*									1.541	2.3
1.4508	30											1.490	4.6
1.3882	20			1.4505	23			1.541	5.7				
1.3740	10			1.3888	23			1.453	1.1				
1.2766	10	1.277	4.9					1.375	4.2				
1.1068	10	1.1062	4.9							1.449	8.7		
1.0768	30	1.0785	1.6*										
1.0755	30			1.0758	23								

**TABLE 14.** Diffraction lines (d) and relative intensities ( $I/I_0$ ) of the sample AUE 37-brown layer compared to lines of the most probable components from  $\mu$ PDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE	CaSO <sub>4</sub> ·2H <sub>2</sub> O (Gypsum) 33-0311*		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		SiO <sub>2</sub> (Quartz) 5-0490D		CaCO <sub>3</sub> (Calcite) 5-0586*		Ca <sub>3</sub> Mg(SiO <sub>4</sub> ) <sub>2</sub> (Merwinite) 35-0591*	
	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$	d [Å]	$I/I_0$
7.6034	10	7.63	54							
4.2941	50	4.283	54		4.26	38			4.3350	1.4
3.3457	90				3.343	110				
3.2942	10								3.3248	1.4
3.1977	10								3.1689	3.2
3.0650	40	3.065	41				3.035		3.0666	0.9
3.0506	100			3.046						
2.8798	50	2.873	24	2.876	6.8*					
2.7998	30	2.789	5.4*	2.795	17				2.8164	0.9
2.7761	30			2.780	49					
2.7402	20	2.732	1.1*	2.744	44					
2.6965	70				46				2.7562	12
2.6750	10	2.685	19						2.6865	46
2.6375	20								2.6712	30
2.6019	10	2.597	3.3	2.608	32				2.6530	21
2.4921	10	2.495	6.0							
2.4569	20	2.452	3.3*	2.451	9.7	2.458	13	2.495	2.5086	0.5
2.4246	20			2.433	3.9*				2.4648	1.4
2.4009	20	2.406	2.2*	2.400	9.7					
2.3761	10									
2.3144	10			2.304	2.9	2.282	13	2.285	2.3206	5.5
2.2801	30			2.282	17				2.2851	3.2
2.1927	10			2.196	5.8				2.1916	0.9
2.0911	10	2.086	14	2.094	3.9			2.095	2.1000	0.5
1.9231	10							1.927		
1.9077	10			1.913	4.9			1.913	1.9111	16
1.8728	10	1.8795	6.5					1.875	1.8787	11

TABLE 14 CONTINUE

SAMPLE	CaSO <sub>4</sub> ·2H <sub>2</sub> O (Gypsum)		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite)		SiO <sub>2</sub> (Quartz)		CaCO <sub>3</sub> (Calcite)		Ca <sub>3</sub> Mg(SiO <sub>4</sub> ) <sub>2</sub> (Merwinite)	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
AUE 37- brown layer	33-0311*		9-0351D		5-0490D		5-0586*		35-0591*	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
	1.8212	20	1.821	2.9*	2.817	19			1.8258	0.5
	1.7857	10	1.791	5.8	1.375	12				
	1.3715	10								
	1.3449	10	1.3440	0.5*						

TABLE 15. Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of the sample AUE 37-undamaged plaster compared to lines of the most probable components from μPDSM computer database. For each compound its mineral name and the number of PDF card are given.

SAMPLE	CaSO <sub>4</sub> ·2H <sub>2</sub> O (Gypsum)		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite)		SiO <sub>2</sub> (Quartz)		CaCO <sub>3</sub> (Calcite)		Ca <sub>1.4</sub> Mg <sub>2</sub> (SiO <sub>4</sub> ) <sub>8</sub> (Bredigite)		Ca <sub>2</sub> Al <sub>1.38</sub> Fe <sub>0.62</sub> O <sub>5</sub>	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
AUE 37- undamaged plaster	33-0311*		33-0302*		5-0490D		24-0027D		36-0399*		42-1469*	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
	7.6002	10									7.2485	22
	7.2767	20			4.26	70			4.19	4*		
	4.2495	40	3.824	8			3.852	43	3.849	9	3.8258	2
	3.8499	10	3.786	8					3.797	6		
	3.7594	10							3.586	2*	3.6332	20
	3.5893	10							3.566	4		
	3.5374	10							3.485	13		
	3.4737	40							3.372	19		
	3.4091	20	3.378	11	3.343	200					3.3819	9
	3.3377	80										
	3.2369	10	3.241	9								
	3.1798	10	3.176	8								
	3.1056	10										
	3.0670	10	3.065	105								

TABLE 15 CONTINUE

SAMPLE	CaSO <sub>4</sub> ·2H <sub>2</sub> O (Gypsum)		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite)		SiO <sub>2</sub> (Quartz)		CaCO <sub>3</sub> (Calcite)		Ca <sub>1.4</sub> Mg <sub>2</sub> (SiO <sub>4</sub> ) <sub>8</sub> (Bredigite)		Ca <sub>2</sub> Al <sub>1.38</sub> Fe <sub>0.62</sub> O <sub>5</sub>	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
AUE 37- undamaged plaster												
	3.0299	100					3.030	148				
	2.9704	20							2.955	11		
	2.8821	60	2.873	63	2.877	33			2.835	30		
	2.8131	30			2.814	35			2.778	11*		
	2.7791	70	2.789	14	2.783	158			2.737	142		33
	2.7508	60			2.745	131			2.723	152		
	2.7280	30	2.732	3*	2.718	47						
	2.6910	70	2.685	49								
	2.6676	30							2.669	190		55
	2.6353	60							2.638	8*		109
	2.6048	60			2.610	66			2.598	13		14
	2.5751	10	2.597	8					2.584	8*		
	2.5493	30	2.534	3*	2.545	14			2.553	9		
	2.4888	20	2.495	15					2.484	8		
	2.4559	20	2.452	8	2.448	19		2.495	2.448	13		
	2.4058	20	2.406	6	2.410	20	2.458	24	2.423	13		3
	2.3813	20			2.403	28			2.344	15		
	2.3581	20										
	2.3288	30			2.323	3*						
	2.3039	10	2.291	1*	2.301	6						
	2.2812	80			2.281	35	2.282	24				
	2.2560	20					2.284	27				
	2.2359	20							2.232	47		13
	2.2125	30	2.219	21					2.213	38		
	2.1884	60			2.189	80	2.237	12				
	2.1661	30			2.165	20			2.176	23		16
	2.1469	20										
	2.1260	10	2.142	3	2.129	11	2.128	18	2.111	13		

TABLE 15 CONTINUE

SAMPLE	CaSO <sub>4</sub> ·2H <sub>2</sub> O (Gypsum) 33-0311*		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 33-0302*		SiO <sub>2</sub> (Quartz) 5-0490D		CaCO <sub>3</sub> (Calcite) 24-0027D		Ca <sub>14</sub> Mg <sub>2</sub> (SiO <sub>4</sub> ) <sub>8</sub> (Bredigite) 36-0399*		Ca <sub>2</sub> Al <sub>1.38</sub> Fe <sub>0.62</sub> O <sub>5</sub>	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
AUE 37- undamaged plaster												
	2.1061	10										
	2.0936	30	2.103	2					2.106	9		
	2.0844	40	2.091	9			2.094	40	2.083	13		
	2.0756	20	2.083	9					2.073	27		
	2.0486	30	2.050	22					2.029	9		27
	2.0290	30	2.027	24					2.024	11		
	2.0187	30	2.020	24	1.980	12						
	1.9836	20	1.982	38					1.924	57		
	1.9234	10	1.9115	9					1.902	23		38
	1.9054	30	1.8979	14								3*
	1.8724	20	1.8795	20								12
	1.8550	10	1.8650	4								
	1.8370	10	1.8441	6					1.838	19		
	1.8166	40	1.8210	5	1.817	34						16
	1.8030	20	1.8018	14	1.801	2						
	1.7851	20	1.7899	11								2
	1.7642	20	1.7657	2*								
	1.7275	10	1.7270	8								7
	1.6694	30			1.672	14			1.670	8		
	1.6643	20	1.6640	8					1.663	8		
	1.6555	10			1.659	6			1.660	4*		
	1.6423	10	1.6456	6					1.649	6		
	1.6379	10							1.639	2		
	1.6236	20	1.6209	13					1.622	8		
	1.6050	20	1.6110	16	1.608	2			1.613	4*		
	1.6008	20	1.6040	17					1.604	4		
	1.5847	10	1.5846	6					1.583	6		
	1.5718	20	1.5738	8					1.576	13		2

TABLE 15 CONTINUE

SAMPLE AUE 37- undamaged plaster	CaSO <sub>4</sub> ·2H <sub>2</sub> O (Gypsum) 33-0311*		Ca <sub>2</sub> SiO <sub>4</sub> (L-arnite) 33-0302*		SiO <sub>2</sub> (Quartz) 5-0490D		CaCO <sub>3</sub> (Calcite) 24-0027D		Ca <sub>14</sub> Mg <sub>2</sub> (SiO <sub>4</sub> ) <sub>8</sub> (Bredigite) 36-0399*		Ca <sub>2</sub> Al <sub>1.38</sub> Fe <sub>0.62</sub> O <sub>5</sub>	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
	1.5579	10									1.5649	16
	1.5532	30										
	1.5414	30			1.541	30						
	1.5322	10	1.5327	3							1.5278	16
	1.4910	20	1.4947	1*							1.4955	8
	1.4840	10										
	1.4614	10	1.4591	4								
	1.4511	20									1.4485	4
	1.4401	20	1.4392	7			1.4405	7			1.4095	4
	1.4031	10	1.4015	3							1.3891	5
	1.3874	10										
	1.3808	20										
	1.3744	40										
	1.3662	20	1.3657	7								
	1.3566	10										
	1.3475	10	1.3440	1*							1.3546	3

TABLE 16. Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of the rest after TG analysis of the sample AUE 37-brown layer compared to lines of the most probable components from μPDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG AUE 37- brown layer	SiO <sub>2</sub> (Quartz) 5-0490D		CaSO <sub>4</sub> (Anhydrite) 6-0226D		Ca <sub>2</sub> SiO <sub>4</sub> (L-arnite) 9-0351D		CaO-MgO-Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> 17-0736O		Ca(Mg,Al)(Si,Al) <sub>2</sub> O <sub>6</sub> (Diopside) 41-1370*		KAISi <sub>3</sub> O <sub>8</sub> (Microcline) 22-0675D	
	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
	7.2648	10										
	4.3949	10										
	4.2849	10	4.26	33					4.434	5.7		
											4.285	50



TABLE 16 CONTINUE

The rest after TG AUE 37- brown layer		$\text{SiO}_2$ (Quartz) 5-0490D		$\text{CaSO}_4$ (Anhydrite) 6-0226D		$\text{Ca}_2\text{SiO}_4$ (Larmitte) 9-0351D		$\text{CaO-MgO-Al}_2\text{O}_3\text{-SiO}_2$ 17-0736O		$\text{Ca(Mg,Al)(Si,Al)}_2\text{O}_6$ (Diopside) 41-1370*		$\text{KAlSi}_3\text{O}_8$ (Microcline) 22-0675D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
3.8772	30	3.87	26	3.827	5.2	3.76	9.8	3.856	6.1				
3.8360	10			3.783	5.2*			3.746	32				
3.7649	20					3.72	20						
3.7322	20												
3.5709	20												
3.5555	20												
3.4965	100	3.498	438										
3.4388	20												
3.3469	50	3.343	95					3.339	5.7*				
3.2894	10					3.27	2.4*	3.220	47				
3.2294	20			3.241	9.1								
3.1820	10			3.175	7.8								
3.1221	10	3.118	18										
3.0670	20												
3.0249	20												
2.9937	30												
2.9651	20							3.09	39			3.005	23
2.9286	20							3.01	9.8			2.979	25
2.8907	20							2.97	2.4			2.946	4.3
2.8907	20							2.94	9.8				
2.8771	20							2.88	49			2.898	23
2.8485	80	2.849	153			2.876	23						
2.8240	20			2.814	13								
2.7917	40	2.797	18	2.795	65								
2.7674	40			2.780	59								
2.7496	50			2.744	62								
2.7191	20			2.731	26							2.774	9.6
2.6945	40												
2.6665	30												

TABLE 16 CONTINUE

The rest after TG AUE 37- brown layer		SiO <sub>2</sub> (Quartz) 5-0490D		CaSO <sub>4</sub> (Anhydrite) 6-0226D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		CaO-MgO-Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> 17-0736O		Ca(Mg,Al)(Si,Al) <sub>2</sub> O <sub>6</sub> (Diopside) 41-1370*		KAlSi <sub>3</sub> O <sub>8</sub> (Microcline) 22-0675D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.6765	40					2.608	42					2.612	7.8
2.6048	20					2.543	9.1	2.550	2.4	2.555	19	2.556	18
2.5444	20							2.521	9.8	2.512	28	2.528	4.3
2.5198	20			2.473	35			2.463	9.8				
2.4712	40					2.451	13	2.452	9.8			2.463	0.9
2.4562	20	2.458	11			2.433	5.2*	2.434	9.8			2.420	4.3
2.4405	20					2.407	13	2.409	9.8			2.401	6.1
2.4052	20												
2.3752	20							2.362	20			2.366	3.5
2.3525	10			2.328	88			2.336	9.8			2.329	5.2
2.3271	80					2.322	3.9*			2.301	13		
2.3022	20					2.304	3.9						
2.2854	10	2.282	11			2.282	23					2.278	1.7
2.2394	10	2.237	5.7					2.245	20	2.228	3.8	2.251	1.7
2.2050	90			2.208	88					2.204	5.7	2.207	1.7
2.1813	50			2.183	35							2.192	2.6
2.1474	30					2.196	7.8					2.160	14
2.1148	10					2.188	42			2.146	15	2.115	7.8
2.1033	30							2.107	20	2.125	17		
2.0851	70			2.086	44					2.102	11		
2.0657	10					2.083	5.2*	2.067	9.8			2.065	2.6
2.0483	30					2.048	13					2.049	3.5
2.0365	30							2.038	39	2.032	11		
2.0251	20					2.026	13	2.027	20				
2.0151	10					2.019	13			2.018	6.6	2.017	3.5
1.9920	40			1.993	26							2.000	4.3
1.9760	10	1.980	5.7			1.983	23	1.975	9.8	1.969	5.7	1.980	4.3
1.9383	40			1.938	18			1.936	39			1.947	1.7

TABLE 16 CONTINUE

The rest after TG AUE 37- brown layer		SiO <sub>2</sub> (Quartz) 5-0490D		CaSO <sub>4</sub> (Anhydrite) 6-0226D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		CaO-MgO-Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> 17-0736O		Ca(Mg,Al)(Si,Al) <sub>2</sub> O <sub>6</sub> (Diopside) 41-1370*		KAlSi <sub>3</sub> O <sub>8</sub> (Microcline) 22-0675D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.9238	30					1.897	3.9					1.928	6.1
1.9041	30												
1.8831	30												
1.8666	80	1.869	70			1.845	3.9						
1.8515	30	1.852	18									1.832	8.5
1.8328	20					1.821	3.9*						
1.8151	50												
1.7468	80	1.749	53									1.742	8.5
1.6468	80	1.648	61										
1.6012	10					1.604	3.9						
1.5926	20	1.594	18										
1.5736	20					1.575	7.8						
1.5624	30	1.564	26										
1.5532	40					1.552	6.5*						
1.5448	10											1.5418	3.8
1.5361	20	1.541	14									1.5291	2.8*
1.5241	40					1.523	5.2*					1.5236	4.7
1.4889	70			1.525	18	1.485	5.2*					1.4911	0.9*
1.4243	40	1.490	26*	1.424	18							1.4192	9.5
1.3963	30	1.424	18	1.398	18	1.397	3.9*					1.4006	1.9
1.3738	30											1.3741	1.9*
1.3464	20	1.375	10										
1.3188	20			1.319	18							1.3256	3.8
1.3035	10												
1.2958	20			1.296	9								
1.2761	50	1.277	26	1.277	26								
1.2150	40	1.216	18	1.216	18							1.2182	0.9
1.2035	10											1.2055	0.9

TABLE 16 CONTINUE

The rest after TG AUE 37- brown layer		SiO <sub>2</sub> (Quartz) 5-0490D		CaSO <sub>4</sub> (Anhydrite) 6-02226D		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 9-0351D		CaO-MgO-Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> 17-0736O		Ca(Mg,Al)(Si,Al) <sub>2</sub> O <sub>6</sub> (Diopside) 41-1370*		KAlSi <sub>3</sub> O <sub>8</sub> (Microcline) 22-0675D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.2012	10	1.1997	4.7	1.1993	9								
1.1781	10	1.1802	3.8	1.1781	4								
1.1655	40			1.1663	18								
1.1482	20			1.1483	9								
1.1052	40			1.1062	26								
1.0779	20			1.0785	9								

TABLE 17. Diffraction lines (d) and relative intensities (I/I<sub>0</sub>) of the rest after TG analysis of the sample AUE 37-undamaged plaster compared to lines of the most probable components from μPDSM computer database. For each compound its mineral name and the number of PDF card are given.

The rest after TG AUE 37- undamaged plaster		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> Al(Al,Si) <sub>2</sub> O <sub>7</sub> (Gehlenite) 25-0123D		Al <sub>2</sub> O <sub>3</sub> (Corundum) 42-1468*		Ca <sub>5</sub> MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub> 13-0272		CaAl <sub>2</sub> SiO <sub>6</sub> 31-0249I		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 24-0037D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
7.2648	10												
4.2605	40	4.26	70	4.24	8.0*					4.33	4.9	4.264	6
3.8303	30			3.72	27							3.827	2*
3.7664	20												
3.5639	20			3.45	1.8*	3.48	41						
3.4945	50												
3.3475	100	3.343	201	3.07	22							3.378	6
3.0841	30							3.039	39			3.048	15
3.0329	40							2.973	15				
2.9747	40			2.857	88					2.94	81		
2.8812	30			2.730	13*			2.776	77	2.904	28	2.878	30
2.7841	90							2.748	31	2.904		2.793	112
2.7516	90											2.748	124
2.6961	90											2.717	123

TABLE 17 CONTINUE

The rest after TG AUE 37- undamaged plaster		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> Al(Al,Si) <sub>2</sub> O <sub>7</sub> (Gehlenite) 25-0123D		Al <sub>2</sub> O <sub>3</sub> (Corundum) 42-1468*		Ca <sub>5</sub> [MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub> 13-0272		CaAl <sub>2</sub> SiO <sub>6</sub> 31-0249I		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 24-0037D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
2.6375	80											2.610	97
2.6085	80					2.551	56	2.608	54	2.535	33	2.582	24
2.5486	50			2.535	13					2.509	24	2.546	7*
2.5185	20			2.441	27					2.478	41	2.503	16
2.4881	10			2.406	27	2.379	24					2.493	33
2.4498	20	2.458	24									2.452	4
2.4061	60												
2.3782	40												
2.3434	30									2.340	3.3*		
2.3302	30							2.323	7.7				
2.2943	20			2.299	18							2.305	22
2.2820	50	2.282	24							2.274	16	2.282	46
2.2487	10	2.237	12										
2.1892	80			2.199	2.7*					2.188	13*	2.189	79
2.1276	30	2.128	18									2.132	4
2.1040	40									2.106	16	2.106	24
2.0844	50					2.085	58			2.090	16	2.084	20
2.0672	10									2.075	9.8	2.065	5
2.0501	30			2.048	8.8							2.047	26
2.0221	30											2.026	7
2.0146	20									2.005	15	2.021	5
1.9834	30	1.980	12					1.983	7.7	1.991	28	1.984	29
1.9411	30							1.938	7.7	1.948	9.8*		
1.9238	20			1.930	49								
1.9048	20											1.894	36
1.8688	10			1.872	7.1								
1.8494	10			1.860	3.5							1.847	7
1.8210	20	1.817	34	1.819	49			1.823	7.7	1.808	4.9	1.821	6

TABLE 17 CONTINUE

The rest after TG AUE 37- undamaged plaster		SiO <sub>2</sub> (Quartz) 5-0490D		Ca <sub>2</sub> Al(Al,Si) <sub>2</sub> O <sub>7</sub> (Gehlenite) 25-0123D		Al <sub>2</sub> O <sub>3</sub> (Corundum) 42-1468*		Ca <sub>5-1</sub> MgAl <sub>2</sub> Si <sub>16</sub> O <sub>90</sub> 13-0272		CaAl <sub>2</sub> SiO <sub>6</sub> 31-0249I		Ca <sub>2</sub> SiO <sub>4</sub> (Larnite) 24-0037D	
d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>	d [Å]	I/I <sub>0</sub>
1.7636	20			1.7608	40			1.766	31			1.762	9
1.6979	10									1.703	13	1.697	12
1.6495	10	1.659	6									1.659	5
1.6003	20	1.608	2*			1.6014	48			1.596	16	1.597	4
1.5529	30												
1.5414	20	1.541	30			1.5459	1.2*	1.541	7.7	1.540	3.3		
1.5225	20												
1.5094	20					1.5109	4.1*			1.512	24		
1.4882	40							1.488	15	1.497	16		
1.4041	30					1.4045	17			1.399	15		
1.3825	30	1.382	14	1.3790	22					1.382	24		
1.3747	30	1.375	22			1.3738	26			1.202	1.6		
1.1997	20	1.1997	10										

FIGURE 3. The Mössbauer spectrum of  $^{57}\text{Fe}$  in sample DDD 36-undamaged plaster underneath the brown layer at room temperature I –  $\text{Fe}^{3+}_{(1)}$  - tetrahedral sites, II –  $\text{Fe}^{3+}_{(2)}$  - octahedral sites

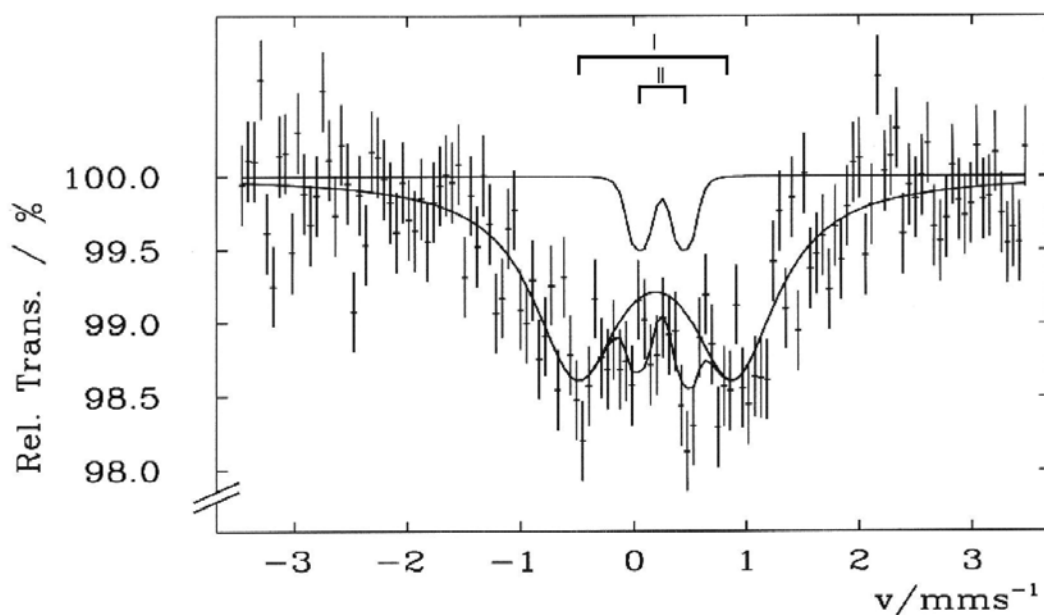
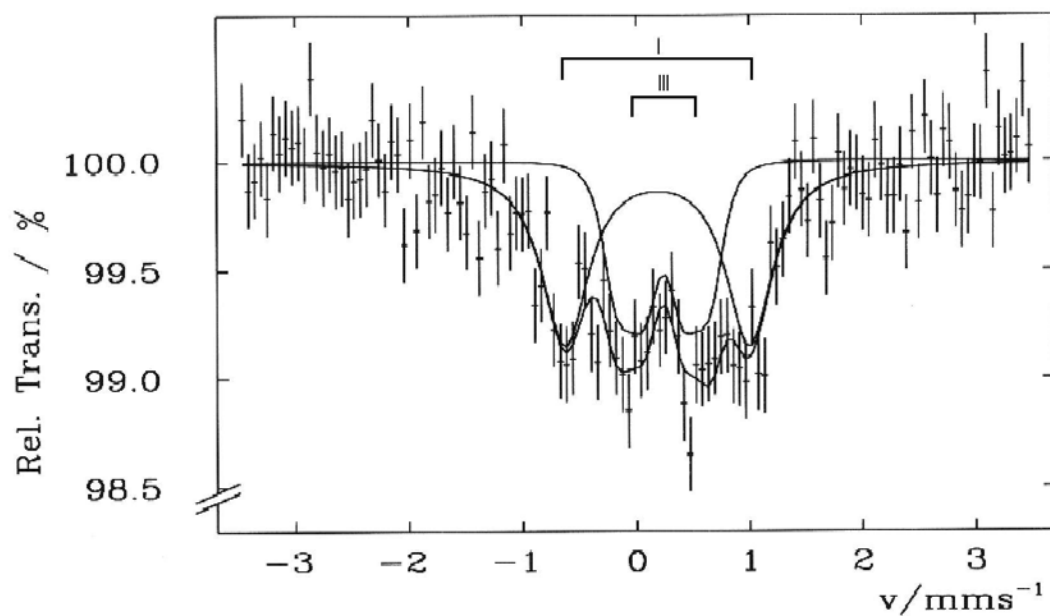


FIGURE 4. The Mössbauer spectrum of  $^{57}\text{Fe}$  in sample DDD 36-brown layer at room temperature I –  $\text{Fe}^{3+}_{(1)}$  - tetrahedral sites, III –  $\text{Fe}^{3+}_{(3)}$  - octahedral sites

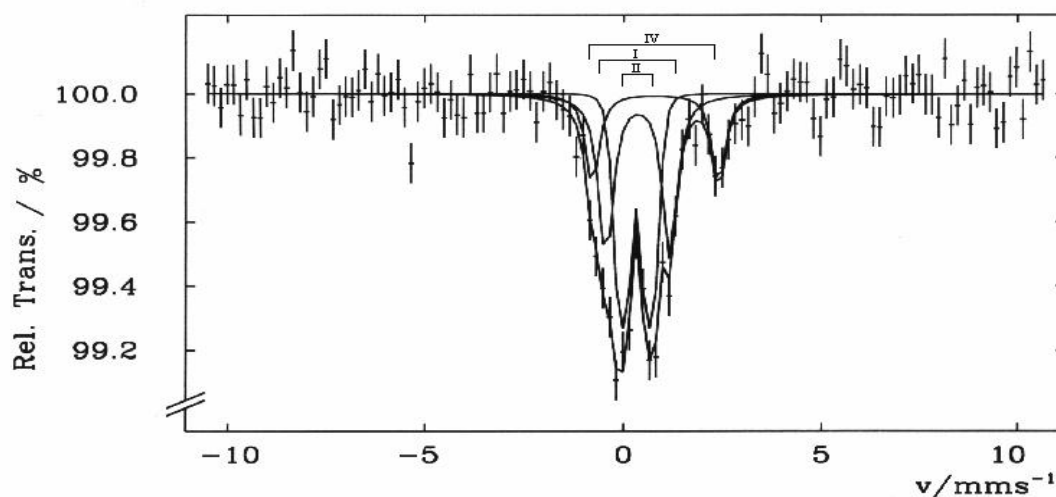


**TABLE 18.** Hyperfine parameters of Mössbauer spectrum of the sample DDD 36 at room temperature

a.) undamaged plaster

b.) brown layer

	oxidation state	surface area	isomer shift $\delta$ [mm/s]	quadrupole splitting $\Delta E_Q$ [mm/s]	line width $\Gamma$ [mm/s]	relative intensity [%]
a.)	$\text{Fe}^{3+}_{(1)}$	$789 \pm 226$	$0.29 \pm 0.07$	$1.39 \pm 0.18$	$0.94 \pm 0.32$	90
	$\text{Fe}^{3+}_{(2)}$	$85 \pm 96$	$0.35 \pm 0.07$	$0.40 \pm 0.11$	$0.26 \pm 0.18$	10
b.)	$\text{Fe}^{3+}_{(1)}$	$722 \pm 217$	$0.28 \pm 0.04$	$1.62 \pm 0.11$	$0.50 \pm 0.16$	51
	$\text{Fe}^{3+}_{(3)}$	$611 \pm 154$	$0.33 \pm 0.04$	$0.59 \pm 0.07$	$0.48 \pm 0.08$	49

**FIGURE 5.** The Mössbauer spectrum of  $^{57}\text{Fe}$  in the sample ZMV 7 – brown layer at room temperature  
I –  $\text{Fe}^{3+}_{(1)}$  - tetrahedral sites, II –  $\text{Fe}^{3+}_{(3)}$  - octahedral sites, IV –  $\text{Fe}^{2+}$ **TABLE 19.** Hyperfine parameters of Mössbauer spectrum of the sample ZMV 7 at room temperature

a.) undamaged plaster

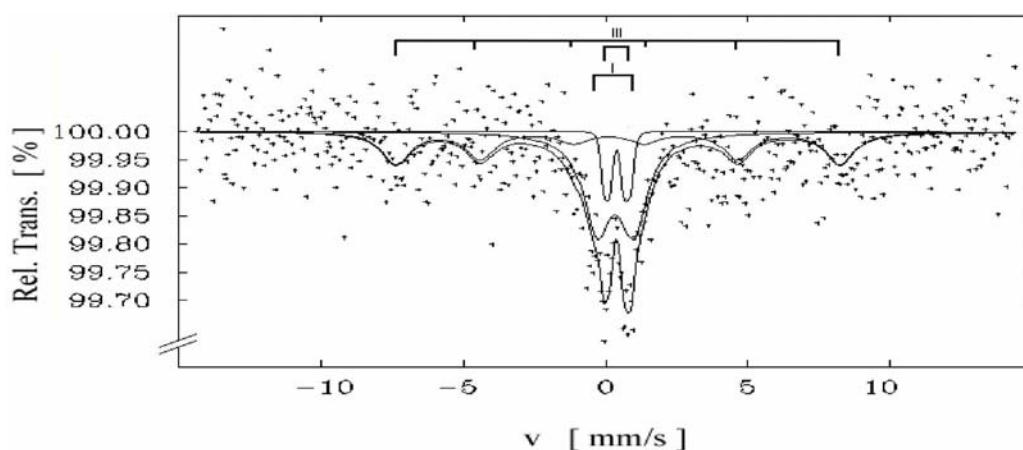
c.) brown layer

	oxidation state	surface area	isomer shift $\delta$ [mm/s]	quadrupole splitting $\Delta E_Q$ [mm/s]	line width $\Gamma$ [mm/s]	relative intensity [%]
a.)	$\text{Fe}^{3+}_{(1)}$	$4595 \pm 1036$	$0.28 \pm 0.05$	$1.44 \pm 0.15$	$0.70 \pm 0.18$	68
	$\text{Fe}^{3+}_{(2)}$	$2152 \pm 803$	$0.34 \pm 0.05$	$0.43 \pm 0.08$	$0.42 \pm 0.10$	32
b.)	$\text{Fe}^{3+}_{(1)}$	$2762 \pm 1037$	$0.40 \pm 0.04$	$1.61 \pm 0.12$	$0.40 \pm 0.16$	31
	$\text{Fe}^{3+}_{(3)}$	$4625 \pm 961$	$0.36 \pm 0.03$	$0.69 \pm 0.07$	$0.50 \pm 0.08$	53
	$\text{Fe}^{2+}$	$1420 \pm 588$	$0.85 \pm 0.06$	$3.18 \pm 0.12$	$0.36 \pm 0.16$	16



From the room temperature Mössbauer spectra it was not possible to distinguish between goethite ( $\alpha$ -FeOOH) and ferrihydrite ( $\text{Fe}_5\text{HO}_8 \cdot 4\text{H}_2\text{O}$ ). It is only obvious that there is a difference in chemical environment of ferric ions in brown layers and in the corresponding undamaged plaster underneath the brown layers. Therefore a low temperature spectrum (Figure 6) at 73 K was recorded for brown layer of sample DDD 36. The spectrum shows a sextet pattern as well as two overlapping doublet patterns. A hyperfine magnetic field appears because some of the particles have reached magnetic order. The linewidths are quite large. The results are shown in Table 20.

**FIGURE 6.** The Mössbauer spectrum of  $^{57}\text{Fe}$  in the sample DDD 36-brown layer at temperature 73 K  
I –  $\text{Fe}^{3+}_{(1)}$  - tetrahedral sites, III –  $\text{Fe}^{3+}_{(3)}$  - octahedral sites



**TABLE 20.** Hyperfine parameters of Mössbauer spectrum of the sample DDD 36-brown layer recorded at

73 K.

- a.) doublet  
b.) sextet

	oxidation state	surface area	isomer shift $\delta$ [mm/s]	quadrupole splitting $\Delta E_Q$ [mm/s]	line width $\Gamma$ [mm/s]	hyperfine magnetic field H [T]
a.)	$\text{Fe}^{3+}_{(1)}$	$3402 \pm 1280$	$0.39 \pm 0.11$	$1.30 \pm 0.61$	$1.18 \pm 0.38$	/
	$\text{Fe}^{3+}_{(3)}$	$958 \pm 1392$	$0.46 \pm 0.06$	$0.72 \pm 0.16$	$0.44 \pm 0.26$	/
b.)	$\text{Fe}^{3+}_{(3)}$	$2653 \pm 1035$	0.33	-0.63	1.12	48.6

### Discussion

Samples of brown layers and samples of corresponding undamaged plaster underneath consist of elements: Ca, Si, Al, Mg, K, S, Fe, Ti, O, C, H.

The DTA endothermic effect at 100 °C (Figure 1) can be ascribed to the desorption of surface bound water. A slow, continuous mass loss in the temperature range of 100-500 °C is a characteristic of OH group polycondensation. A significant mass loss in the temperature range of 600-900 °C belongs to the degradation of carbonates.<sup>7</sup> In samples DDD 36-undamaged plaster (Figure 2) and ZMV 7-undamaged plaster are present both CaCO<sub>3</sub> and CaMg(CO<sub>3</sub>)<sub>2</sub>. On account of that the degradation of carbonates occurs in two steps. CaMg(CO<sub>3</sub>)<sub>2</sub> decomposes in the temperature range of 400-700 °C.<sup>7</sup>

The IR spectra confirm conclusions from the TG analysis. A broad band (sample DDD 36-brown layer) in the range of 3700-3000 cm<sup>-1</sup> can be assigned to vibrations of water.<sup>8</sup> Since the band is broad it can be assumed that molecules of water are differently bound to metals present in the sample. In the residual after TG analysis broad band in this area disappears and the sharp band at 3640 cm<sup>-1</sup> can be ascribed to stretching mode of O-H group.<sup>8</sup> It appears due to partial hydration of metal oxides on air after TG analysis. Strong, broad band in the range of 1500-1400 cm<sup>-1</sup> and two sharp peaks at 873 cm<sup>-1</sup> and 713 cm<sup>-1</sup> belong to -CO<sub>3</sub> modes.<sup>8</sup> The first one is significant for ν<sub>3</sub>(C-O) stretching modes, the second one can be ascribed to π modes of the whole carbonate group and the last one to ν<sub>4</sub>(O-C-O) bending modes. These bands are missing in the spectrum taken on the rest after TG analysis. The broad band in the range of 1200 – 800 cm<sup>-1</sup> appears in both spectra and can be ascribed to M – O modes<sup>8</sup> of various metal oxides present in samples. Because IR spectra didn't give the information which metal oxides are present, we didn't record spectra of all samples.

Components which show the best matching of lines in X-ray powder patterns are CaCO<sub>3</sub>, SiO<sub>2</sub>, Ca<sub>2</sub>SiO<sub>4</sub>, CaMg(CO<sub>3</sub>)<sub>2</sub>, silicate minerals containing Ca, Al, Mg, K, Fe, metal hydroxides and sulphates. It was possible to determine more components in

residuals after TG analysis. In addition to better crystallised phases, e.g. SiO<sub>2</sub>, a number of new phases appeared in residuals due to chemical reactions.

From the above analyses it could not be concluded that iron is involved in chemical changes of external layers of building facades. Mössbauer analysis turned out to be a convenient method for investigation of iron in poorly crystalline material.

Two overlapping doublet patterns (Figure 3) in the room temperature spectra of undamaged plaster of the sample DDD 36 can be ascribed to ferric ions on tetrahedral ( $\text{Fe}^{3+}_{(1)}$ ) and octahedral ( $\text{Fe}^{3+}_{(2)}$ ) sites in ferrites of brownmillerite-like structure<sup>2</sup>. In brown layers coordination remains tetrahedral and octahedral ( $\text{Fe}^{3+}_{(3)}$ ), but there are more ferric ions at octahedral sites what can be concluded by comparing relative intensities (Table 18 and 19). The change in quadrupole splitting parameter of about 0.2 mm/s indicates a change in chemical environment of the octahedrally coordinated ferric ions ( $\text{Fe}^{3+}_{(3)}$ ). Parameters of  $\text{Fe}^{3+}_{(3)}$  agree well with parameters of microcrystalline goethite ( $\alpha\text{-FeOOH}$ )<sup>9</sup> and ferrihydrite ( $\text{Fe}_5\text{HO}_8\cdot 4\text{H}_2\text{O}$ ).<sup>3,9</sup> To determine which one of these two minerals is present in brown layers the low temperature spectrum at 73 K was recorded (Figure 6). Parameters of this spectrum (Table 20) match with parameters of microcrystalline goethite.<sup>9</sup> The hyperfine magnetic field is 1 T lower than that found in literature, which can be explained by aluminium substitution, sample crystallinity and particle size effects.<sup>10,11</sup> The difference between brown layers and undamaged plaster underneath the brown layers is in formation of goethite ( $\alpha\text{-FeOOH}$ ). It is microcrystalline and some of the  $\text{Fe}^{3+}$  ions are substituted by a non magnetic ion, probably  $\text{Al}^{3+}$ .

Iron in oxidation state 2+ is present only in the brown layer of sample ZMV 7 (Figure 5). One of the possible ways of its genesis is reduction of  $\text{Fe}^{3+}$  due to SO<sub>2</sub> present in the air.

### Acknowledgements

The authors wish to thank to the Ministry of Education, Science and Sport of the Republic of Slovenia for their financial support (Grant PO – 0508-0103).

### References and Notes

1. I. Nemeč, *Društveni dom Domžale – barvna študija*, Konzervatorski nadzor: Zavod za varstvo naravne in kulturne dediščine, Kranj 1995
2. D. Hanžel, G. Lahajnar, *A Study of Two Yugoslav Cements by Mössbauer Spectroscopy of  $^{57}\text{Fe}$* , Vestnik slovenskega kemijskega društva, Ljubljana, 1986, 33, 147-150
3. A. J. Nord and T. Ericson, Chemical Analysis of Thin Black Layers on Building Stone, *Studies in Conservation*, 1993, 38, 25-35
4. W. E. Steger and H. Mehner, *The Iron in Black Weathering Crusts on Saxonian Sandstones Investigated by Mössbauer Spectroscopy*, *Studies in Conservation*, 1998, 43, 49-58
5. PDF, Sets 1-49 and 70-86, International Centre for Diffraction Data, Newtown Square, Pennsylvania, USA 1999
6.  $\mu\text{PDSM}$ , "Micro Powder Diffraction Search Match", Fein-Marquart Associates. Release 4.30
7. *Atlas of Thermoanalytical Curves*, ed. G. L. Pyay, Akadémiai Kiadó, Budapest 1977
8. K. Nakamoto, *Infrared and Raman Spectra*, Wiley & Sons, 4<sup>th</sup> Edition, 1986
9. E. Murad and J. H. Johnston, *Iron Oxides and Oxyhydroxydes*, Mössbauer Spectroscopy Applied to Inorganic Chemistry, Vol. 2, ed. G. J. Long, Plenum Press, New York 1987
10. S. Mørup, *Mössbauer Studies of Microcrystalline Materials*, Mössbauer Spectroscopy Applied to Inorganic Chemistry, Vol. 2, ed. G. J. Long, Plenum Press, New York 1987
11. D. C. Golden, L. H. Bowen, S. B. Weed and J. M. Bingham, *Mössbauer Studies of Synthetic and Soil – occurring Aluminium – Substituted Goethites*, *Soil Sci. Am. J.*, 1979, 43, 802-808

### Povzetek

Na nekaterih arheoloških stavbah je fasada tik pod površino rjavo obarvana. Ta rjava obarvanost je posledica kemijskih sprememb, ki lahko pozročijo tudi nadaljno destrukcijo fasade.

Rjavo plast in nepoškodovano plast fasade tik pod rjavo plastjo sestavljajo  $\text{CaCO}_3$ ,  $\text{SiO}_2$ ,  $\text{Ca}_2\text{SiO}_4$ ,  $\text{CaMg}(\text{CO}_3)_2$ , silikati, ki vsebujejo Ca, Al, Mg, K, Fe, kovinski hidroksidi in sulfati. V obeh plasteh je tudi nekaj površinsko vezane vode.  $\text{Fe}^{3+}$  ioni so tetraedrično in oktaedrično koordinirani. V rjavi plasti je več  $\text{Fe}^{3+}$  ionov na oktaedričnih mestih in eksperimentalni podatki kažejo tudi na spremembo kemijskega okolja  $\text{Fe}^{3+}$  ionov. Razlika med rjavo plastjo in nepoškodovano plastjo fasade je v tem, da je v rjavi plasti nastal goethit ( $\alpha$  -  $\text{FeOOH}$ ), ki je mikrokristaliničen in delež  $\text{Fe}^{3+}$  ionov je substituiran z ne – magnetnim ionom, verjetno  $\text{Al}^{3+}$  ionom. Železo v oksidacijskem stanju 2+ je prisotno samo v rjavi plasti vzorca ZMV 7. Ena od možnih poti nastanka je z redukcijo  $\text{Fe}^{3+}$ , zaradi prisotnosti  $\text{SO}_2$  v zraku.