Petrogeneza in uporabna vrednost lojevčevih kamnin na območjih Wonu-Apomu in Ilesa v jugozahodni Nigeriji

Anthony T. Bolarinwa^{1,*}, Morenike A. Adeleye¹

¹University of Ibadan, Department of Geology, Ibadan, Nigeria *Corresponding author. E-mail: atbola@yahoo.com

Abstract

Petrogenesis and functional applications of the talcose rocks in Wonu-Apomu and Ilesa areas, southwestern Nigeria, were undertaken. Petrographic studies of the talcose rocks showed talc with variable amounts of anthophyllite, and chlorite. X-ray diffraction studies of the talcose rocks further revealed the presence of Al-bearing pyrophyllite (19.0 %) and Cr-bearing clinochlore (14.5 %), which have similar physical characteristics and diffraction peaks with talc and chlorite, respectively thus increasing the intensity of talc and chlorite on the x-ray diffraction charts.

Chemical data of the talcose rocks showed that the Wonu-Apomu samples are more siliceous (ca. 55.62 % SiO₂) than the Ilesa samples with ca. 52.35 %. Trace element data showed higher Cr (> 3 600 µg/g), Ni (> 1 620 µg/g) and Zn (> 160 µg/g) contents in the Ilesa talcose rocks. Petrogenetic indices, including low K, Rb, Rb/Sr (< 0.29), K/Rb (< 49.55), Ni/Co (< 1.05), Ga/Y (< 1.05) and CaO/Al₂O₃ (< 0.72) suggested tholeiitic basalt precursor for the talc bodies. The composition, physical and industrial characteristics of the talc bodies supported functional applications as fillers, filters and absorbents in ceramics, paints, rubber, paper, plastic, roofing materials and textiles.

Key words: talcose rock, mineralogy, geochemistry, petrogenesis, functional applications

Izvleček

Predmet preiskave sta bili določitev petrogenetskih značilnosti in ocena uporabne vrednosti lojevec vsebujočih kamnin s področij Wonu-Apomu in Ilesa v jugozahodni Nigeriji. S petrografskimi preiskavami kamnin so ugotovili razen lojevca še različne deleže antofilita in klorita. Rentgenske difrakcijske preiskave kažejo na prisotnost Al-vsebujočega pirofilita (19,0 %) in Cr-vsebujočega klinoklora (14,5 %), ki imata podobne fizikalne lastnosti in difrakcijske vrhove kot lojevec, iz česar izhaja zvečana intenziteta lojevca in klorita na rentgenskih difrakcijskih diagramih.

Iz podatkov kemijske analize je mogoče sklepati, da vsebujejo vzorci iz Wonu-Apomuja več kremena (povpr. 55,62 % SiO₂) od vzorcev iz Ilese, ki ga imajo povprečno 52,35 %. Analize slednih prvin kažejo v prvih več Cr (> 3 600 µg/g), Ni (> 1 620 µg/g) in Zn (> 160 µg/g) kakor v vzorcih lojevčevih kamnin iz Ilese. Petrogenetski kazalci, med drugim nizek K, Rb, Rb/Sr (< 0,29), K/Rb (< 49,55), Ni/Co (< 1,05), Ga/Y (< 1,05) in CaO/Al₂O₃ (< 0,72), nakazujejo, da je bila izvirna kamnina lojevca tholeiitni bazalt. Glede na sestavo lojevčevih teles in njihove fizikalne ter industrijske značilnosti ugotavljajo, da je mogoče kamnino uporabljati kot polnilo, material za filtre in kot absorbent, v keramiki, izdelavi barvil, gume, papirja, plastičnih snovi, kritine in v tekstilni industriji.

Ključne besede: lojevčeva kamnina, mineralogija, geokemija, petrogeneza, uporabna vrednost

Introduction

Talc bearing rocks and their protolithic amphibolites are common features of the Precambrian schist belts, which are largely confined to the western part of Nigeria (Figure 1). In the Ilesa schist belt of southwestern Nigeria, they occur in localities such as Wonu-Apomu, Ile-Ife, Isaobi near Ilesa. Ikirun and Esa-Oke. Within the Ilesa schist belt, talc bodies are closely associated with mafic and ultramafic bodies, mostly amphibolites and metasedimentary units such as quartzites and pelitic schists ^[1-7]. Various mineralogical and textural varieties of the talcose rocks have been identified by workers such as ^[8-10]. The four major mineralogical types according to Elueze and Akin-Ojo^[10] are the talcose, tremolitic, chloritic and anthophyllitic varieties. Furthermore, studies on the regional occurrences and comparative industrial applications of various talc occurrences in southwestern Nigeria have also been carried out ^[10, 11].

Industrial appraisals of various talc bodies from southwestern Nigeria have also been inferred from their textural, mineralogical, chemical and geotechnical attributes ^[11-18]. Most of these investigations on the amphibolites and various talcose occurrences within the Precambrian basement complex tend to emphasize their field relations, petrological descriptions,

Prh lobeti ellorin BENIN Egbe Kabba Loko 5 Ado-Ekiti • Okene REPUBLIC Benin-City BIGHT -BENIN OF Cretaceous to Recent Sediments Complex Precombrian Basement (Undifferentiated) Pcb Study area 0 20 40 60 80 100 km

Figure 1: Map of southwestern Nigeria showing the location of Apomu and Ilesha areas.

mineralogical attributes and aspects of industrial applications, with little effort on the petrogenetic affinity. Mineralogical and chemical data were generated in the present investigation to unravel the petrogenetic affinity and petrochemical trends of the talcose rocks in Wonu-Apomu and Ilesa areas. The functional applications were also evaluated based on physical and thermal characteristics of the talcose bodies.

Location and Geology of study areas

The Wonu-Apomu area is delineated by latitude 7° 15′ and 7° 19′ N and longitude 4° 3′ and 4° 6′ E (Figure 2); while the Ilesa area is defined by latitude 7° 31′ and 7° 38′ N and longitude 4° 38′ and 4° 45′ E (Figure 3). The Precambrian basement complex rocks in Nigeria have been classified into three major groups. These are the ancient migmatite-gneiss-quartzite com-

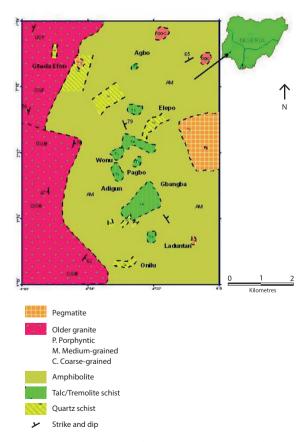


Figure 2: Geological map of Wonu-Apomu area (modified after Akin-Ojo⁽⁹⁾).

plex, the schist belts and the Pan African intrusive series (Older Granites). Others minor rocks include unmetamorphosed felsic and mafic intrusives ^[19].

The Wonu-Apomu and Ilesa areas lie within the schist belts of the basement complex of southwestern Nigeria characterized by migmatites and granitic gneiss, low to medium grade metasedimentary and metavolcanics rocks, notably quartzite and quartz schist, amphibolites and talc schist. Others are porphyritic granite and pegmatites (Figures 2 and 3). Talc schist occurs as narrow, northerly trending, and lensshaped discontinuous bodies within the amphibolite at Pagbo, Wonu, Baale and Laduntan in the Wonu-Apomu area (Figure 4). In Ilesa area, the talc schists occur within biotite schist and amphibolitic rocks. They contain essentially talc with subordinate amounts of anthophyllite, pyrophyllite and chlorite. Chalcopyrite, pyrite, pyrrhotite and chromite, though intensively

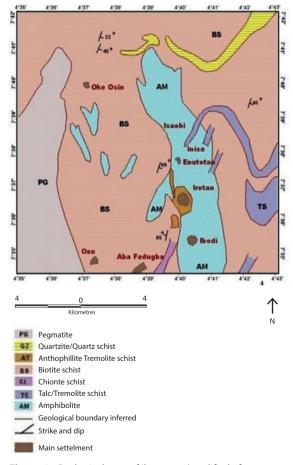


Figure 3: Geological map of Ilesa area (modified after Kehinde-Phillips^[7]).



Figure 4: Talc schist outcrop in Wonu-Apomu area.

altered, are also present in some samples ^[20]. The mineralogy, feel, colour, texture, area extent, and the degree of weathering of the talcose rocks vary with location. The chlorite content of the talc-schist is reflected in its greenish colour, while the weathering effect is indicated by brownish colouration due to iron oxidation.

Materials and Methods

Forty samples of talc schist were collected from outcrops in Wonu-Apomu and Ilesa areas. The samples were selected for thin sectioning, X-ray diffraction studies and chemical analysis. Thin section of talc-schist and amphibolite samples were prepared and examined under petrographic microscope. X-ray diffraction (XRD) study of the talc schist samples were carried out in order to identify mineral phases that could not be identified with optical characteristics. Ten representative samples were selected from Wonu-Apomu and Ilesa areas. The samples were pulverized, pressed into an aluminium sample holder and analysed using Panalytical X' pert Pro diffractometer. Diffraction peaks obtained in 2θ degrees and nm-values were compared with established standards and interpreted with reference to the Joint Committee on Powder Diffraction Standards, Tables of X-ray powder diffraction patterns ^[21].

Twenty (20) talc samples were pulverized into fine powder and analysed at the Activation Laboratory in Ontario, Canada using the Inductively Coupled Plasma-Mass Spectrometry (ICP-MS). The ICP-MS technique was used for the determination of major and trace element composition of the talc bodies, which include SiO_2 , Al_2O_3 , $Fe_2O_3(T)$, MnO, MgO, CaO, Na₂O, K₂O, TiO₂, P₂O₅, Ba, Sr, Y, Zr, Zn, Rb, Co, Cu, V, Ga, Cr and Ni.

Pellets of raw pulverized talc were produced using a mechanical press. The pellets were dried in an oven set at 105 °C and the natural moisture content (NMC) determined. The pellets were fired in a kiln to temperatures of (950, 1 000, 1 050 and 1 100) °C for 2 h. The loss on ignition (LOI) was determined from the weight difference of the dried and fired samples. The linear shrinkage (LSK) was measured and calculated from the percentage decrease in diameter of the pellets after firing. The water absorption capacity was estimated from percentage weight increase after immersion in water for 24 h. Bleaching test for colour improvement was carried out on 10 samples by soaking 2 g in 5 ml of 1.0 M and 2.0 M HCl for 24 h.

Results and Discussion

Petrography

Thin sections of the talcose rock show flaky aggregates and plates of talc (Figures 5–7). Laths of anthophyllite and chlorite are also observed (Figure 5). Aggregates of prismatic needles and radiating fibres of anthophyllites and pyrophyllite were observed in the thin section of the talc schist. The talc-pyrophyllite-chlorite Wonu-Apomu samples are generally greyish to cream white in colour while the anthophyllite-pyrophyllite-chlorite rich types are brownish to grey. Anthophyllite is pleochroic from colourless to yellowish and green in thin section. Pyrophyllite, anthophyllite, chlorite and clinochlore are commonly present in both talcose bodies. A greenish or brownish tint in colour is observed with increase in chlorite and/or anthophyllite content. The pyrophyllite rich varieties are schistose in texture and cream-white in colour. The talc sample has a strong soapy feel. In thin sections, the talc schists are observed as flaky aggregates and platelets of talc (Figures 5–7). Pyrophyllite $(Al_2Si_4O_{10}{OH}_2)$ occurs as fibrous aggregates and rosettes around fine-grained matrix of flaky talc. The pyrophyllite crystals have properties similar to those of talc and are observed as foliated masses in the talc schist. Laths of anthophyllite are also present with subordinate chlorite and clinochlore.

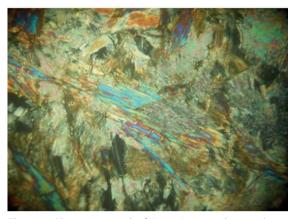


Figure 5: Photomicrograph of Wonu-Apomu talcose rock showing talc and anthophyllite crystals (40-times, crossed polars).



Figure 6: Photomicrograph of Wonu-Apomu talcose rock showing plates of talc (T) (40-times, crossed polars).



Figure 7: Photomicrograph showing plates of talc in the llesa talcose rock (40-times, crossed polars).

X-ray diffraction

X-ray diffraction studies revealed the presence of talc, anthophylite and substantial quantity of pyrophyllite and clinochlore (Table 1). Strong peak of pyrophyllite is close to those of talc and anthophyllite. The chromium bearing clinochlore has similar reflection peaks with chlorite. Talc was identified at (9.5, 28.6, 48.8, and 59.4) 2θ values (Figs. 8 and 9). Peaks of pyrophyllite occur closely with those of talc and anthophyllite at (9.5, 19.4 28.9, 29.4) 2θ values. Anthophyllite peaks are recorded at values of 2θ (9.6, 10.7, 19.8, 27.6, 29.1 and 31.2).

Table 1: Mineral composition (%) of talc schist from

 Wonu-Apomu and Ilesa areas

| Minerals | 1 | 2 | 3 | 4 | Mean |
|---------------|-----|-----|-----|-----|-------|
| Talc | 30 | 32 | 21 | 24 | 26.75 |
| Pyrophyllite | 20 | 20 | 18 | 18 | 19.0 |
| Anthophyllite | 21 | 18 | 28 | 22 | 22.25 |
| Chlorite | 14 | 15 | 16 | 20 | 16.25 |
| Clinochlore | 14 | 14 | 16 | 14 | 14.5 |
| Others | 1 | 1 | 1 | 2 | 1.25 |
| Total | 100 | 100 | 100 | 100 | 100 |
| | | | | | |

1-2: Wonu-Apomu Talc-chlorite-anthophyllite schist

3-4: Ilesa Talc-anthophyllite-chlorite schist

Chlorite in association with clinochlore, a chromium-bearing monoclinic chlorite was identified at 2θ values of (6.2, 12.4, 18.6, 25.0 and 31.2). Chlorite is a hydrous silicate of aluminiun, iron and magnesium (Mg, Fe)₅Al(AlSi₃) O_{10} (OH)₉ while clinochlore with the chemical composition (Mg, Fe, Al)₆ (Si,Cr) O_{10} (OH)₈ is a monoclinic crystal that is distinctly biaxial and optically positive as observed under the petrological microscope. The average clinochlore composition was about 14 % in each of the ten samples analysed. The total chlorite content (including clinochlore) of the talc schist samples was about 30 %, which was responsible for the greenish colour of most of the samples. The mineralogical data of the whole rock samples, using the peak height ratio showed that they are composed of about 30 % talc, 20 % pyrophyllite, 21 % anthophyllite, 28 % chlorite and 1% unidentified minerals.

Comparative studies of the Ilesa and Wonu-Apomu talc-schist with those of Oke-Ila, Iseyin, Baba-Ode and Erin-Omu (Table 2) showed that the Ilesa and Wonu-Apomu samples are higher in pyrophyllite and anthophyllite but lower in talc and tremolite/actinolite contents. The range values of talc for Ilesa and Wonu-Apomu samples is 21–32 % while those of Oke-Ila is about (58–81 %) and Iseyin (48–59 %) ^[15].

Table 2: Comparison of mineral composition (%) of the llesa and Apomu talc schist with some other talc bodies in southwestern

 Nigeria

| | | This | study | | | Oke-Ila | a | Iseyin (TTSC) | | Baba | Odo | Erin-Omu | |
|--------------------------|-----|------|-------|-----|-----|---------|-----|------------------|-----|------|------|--------------|----|
| Minerals | 1 | 2 | 3 | 4 | 2 m | 3 m | 4 m | | | DaDa | -oue | El III-Olliu | |
| Talc | 30 | 32 | 21 | 24 | 58 | 66 | 81 | 59 | 52 | 48 | 24 | 74 | 59 |
| Pyrophyllite | 20 | 20 | 18 | 18 | - | _ | - | - | _ | - | - | _ | - |
| Anthophyllite | 21 | 18 | 28 | 22 | - | - | - | - | - | - | 9 | 3 | 6 |
| Tremolite/ Actinolite | _ | _ | _ | _ | 18 | 20 | 10 | 28 | 26 | 33 | 48 | 7 | 24 |
| Muscovite | - | - | - | - | - | - | - | - | - | - | - | 6 | 2 |
| Chlorite | 14 | 15 | 16 | 20 | 22 | 12 | 8 | 12 | 21 | 17 | 16 | _ | 7 |
| Clinochlore | 14 | 14 | 16 | 14 | - | _ | _ | - | _ | - | - | - | _ |
| Others | 1 | 1 | 1 | 2 | 2 | 2 | 1 | 1 | 1 | 2 | 1 | 1 | 2 |
| Total | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 | | | |

1–2: Wonu-Apomu talc-chlorite-anthophyllite schist 3–4: Ilesa talc-anthophyllite-chlorite schist Oke-Ila: talc-tremolite chlorite schist^[15] TTSC: talc-tremolite schist, Iseyin^[14] Baba-Ode: talc-tremolite chlorite anthophyllite schist [16] Erin-Omu: talc-tremolite chlorite anthophyllite schist [18]

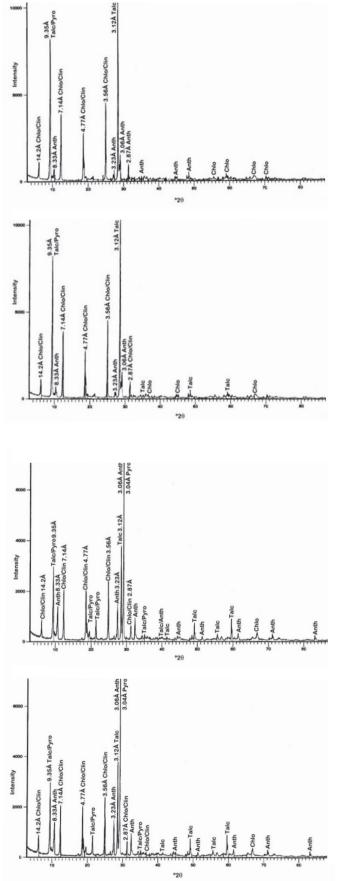
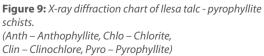


Figure 8: X-ray diffraction charts of Wonu-Apomu talc - chlorite schists. (Anth – Anthophyllite, Chlo – Chlorite, Clin – Clinochlore, Pyro – Pyrophyllite)



| Al.O. 4.16 4.3 3.62 0.84 0.77 5.2 4.99 5.22 5.3 5. Fe ₂ O ₃₀₀ 8.34 7.83 8.27 3.94 3.75 9.54 9.26 9.38 9.42 9.33 MnO 0.14 0.13 0.15 0.48 0.05 0.15 0.14 0.14 0.14 0.14 MgO 28.59 27.89 29.18 30.04 30.06 26.54 26.67 26.03 26.46 26.07 Ga 0.77 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.14 0.11 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0. | | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 |
|---|---|-------------|-------|--------|--------|--------|--------|--------|-------|--------|-------|
| Fe ₂ O ₃₀₀ 8.34 7.83 8.27 3.94 3.75 9.54 9.26 9.38 9.42 9.2 MnO 0.14 0.13 0.15 0.48 0.05 0.15 0.14 0.14 0.14 0.1 MgO 28.59 27.89 29.18 30.04 30.06 26.54 26.67 26.03 26.46 26.03 CaO 2.7 3.17 2.59 0.06 0.05 1.39 1.34 1.38 1.31 1.31 Na ₂ O 0.04 0.01 0.11 0.01 0.02 0.03 0.01 0.01 0.01 0.01 0.01 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.03 0.04 0.04 0.04 0.03 0.02 0.03 0.03 0.03 0.041 99.7 Trace element (µg/g) Ba 32 37 9 10 15 1 | SiO ₂ | 52.29 | 51.43 | 52.23 | 61.15 | 61.01 | 52.42 | 52.82 | 52.1 | 52.24 | 52.15 |
| MnO 0.14 0.13 0.15 0.48 0.05 0.15 0.14 0.14 0.14 0.14 MgO 28.59 27.89 29.18 30.04 30.06 26.54 26.67 26.03 26.46 26.01 CaO 2.7 3.17 2.59 0.06 0.05 1.39 1.34 1.38 1.31 1.31 Na ₂ O 0.04 0.01 0.11 0.03 0.03 0.08 0.07 0.01 0.01 0.01 KgO 0.06 0.05 0.11 0.01 0.02 0.03 0.01 0.01 0.01 0.01 0.03 0.02 0.03 0.03 0.04 0.01 0.11 0.11 0.11 0.11 0.11 0.01 0.01 0.03 0.02 0.03 0.03 0.03 0.03 0.03 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.0 | Al ₂ O ₃ | 4.16 | 4.3 | 3.62 | 0.84 | 0.77 | 5.2 | 4.99 | 5.22 | 5.3 | 5.14 |
| Mg0 28.59 27.89 29.18 30.04 30.06 26.54 26.67 26.03 26.46 26.67 CaO 2.7 3.17 2.59 0.06 0.05 1.39 1.34 1.38 1.31 1.38 Na ₂ O 0.04 0.01 0.11 0.03 0.03 0.08 0.07 0.01 0.01 0.01 K ₂ O 0.06 0.05 0.11 0.01 0.02 0.03 0.01 0.01 0.01 0.01 0.01 0.02 0.03 0.03 0.03 0.03 0.02 0.03 0.03 0.03 0.03 0.03 0.03 0.02 0.03 < | Fe ₂ O _{3(t)} | 8.34 | 7.83 | 8.27 | 3.94 | 3.75 | 9.54 | 9.26 | 9.38 | 9.42 | 9.33 |
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| Na ₂ O 0.04 0.01 0.11 0.03 0.03 0.08 0.07 0.01 0.01 0.01 K ₂ O 0.06 0.05 0.11 0.01 0.02 0.03 0.01 0.08 0.06 0.01 TiO ₂ 0.19 0.20 0.187 0.2 0.02 0.09 0.1 0.1 0.11 0.0 P ₂ O ₅ 0.03 0.03 0.03 0.01 0.01 0.03 0.02 0.03 0.03 0.03 LOI 4.42 4.65 4.21 4.81 4.79 5.22 5.24 5.33 5.27 55 Total 100.96 99.7 100.59 101.57 100.69 100.66 99.8 100.41 99.7 Trace element (µg/g) E E E E E E E Ba 32 30 37 9 10 15 14 18 15 15 Co 69 | MgO | 28.59 | 27.89 | 29.18 | 30.04 | 30.06 | 26.54 | 26.67 | 26.03 | 26.46 | 26.02 |
| k ₂ O 0.06 0.05 0.11 0.01 0.02 0.03 0.01 0.08 0.06 0.01 TiO ₂ 0.19 0.20 0.187 0.2 0.02 0.09 0.1 0.1 0.11 0.1 P ₂ O ₅ 0.03 0.03 0.03 0.01 0.01 0.03 0.02 0.03 0.03 0.03 0.03 LOI 4.42 4.65 4.21 4.81 4.79 5.22 5.24 5.33 5.27 5.5 Total 100.96 99.7 100.59 101.57 100.56 100.69 100.66 99.8 100.41 99.7 Trace element (µg/g) Ba 32 30 37 9 10 15 14 18 15 Co 69 69 70 71 73 72 70 69 70 60 Cu 70 60 70 10 10 20 10 10 10< | CaO | 2.7 | 3.17 | 2.59 | 0.06 | 0.05 | 1.39 | 1.34 | 1.38 | 1.31 | 1.38 |
| TiO2 0.19 0.20 0.187 0.2 0.02 0.09 0.1 0.1 0.11 0.01 P205 0.03 0.03 0.03 0.01 0.01 0.03 0.02 0.03 0.03 0.03 LOI 4.42 4.65 4.21 4.81 4.79 5.22 5.24 5.33 5.27 55 Total 100.96 99.7 100.59 101.57 100.56 100.69 100.66 99.8 100.41 99.7 Trace element (µg/g) E <th< th=""><td>Na₂0</td><td>0.04</td><td>0.01</td><td>0.11</td><td>0.03</td><td>0.03</td><td>0.08</td><td>0.07</td><td>0.01</td><td>0.01</td><td>0.01</td></th<> | Na ₂ 0 | 0.04 | 0.01 | 0.11 | 0.03 | 0.03 | 0.08 | 0.07 | 0.01 | 0.01 | 0.01 |
| P205 0.03 0.03 0.01 0.01 0.03 0.02 0.03 0.03 0.03 LOI 4.42 4.65 4.21 4.81 4.79 5.22 5.24 5.33 5.27 5 Total 100.96 99.7 100.59 101.57 100.56 100.69 100.66 99.8 100.41 99.7 Trace element (µg/g) Ba 32 30 37 9 10 15 14 18 15 7 Go 69 69 70 71 73 72 70 69 70 70 Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 36 Cu 70 60 70 10 10 20 10 10 10 Ga 7 6 5 2 2 8 8 8 Ni 1110 1080 1150< | K ₂ 0 | 0.06 | 0.05 | 0.11 | 0.01 | 0.02 | 0.03 | 0.01 | 0.08 | 0.06 | 0.01 |
| LOI 4.42 4.65 4.21 4.81 4.79 5.22 5.24 5.33 5.27 55 Total 100.96 99.7 100.59 101.57 100.56 100.69 100.66 99.8 100.41 99.7 Ba 32 30 37 9 10 15 14 18 15 16 Co 69 69 70 71 73 72 70 69 70 70 Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 36 Cu 70 60 70 10 10 20 10 10 10 Ga 7 6 5 2 2 8 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 166 Rb 29 6 4 3 2 2 1 3 4 4 4 V | TiO ₂ | 0.19 | 0.20 | 0.187 | 0.2 | 0.02 | 0.09 | 0.1 | 0.1 | 0.11 | 0.11 |
| Total 100.96 99.7 100.59 101.57 100.56 100.69 100.66 99.8 100.41 99.7 Trace element (µg/g) Ba 32 30 37 9 10 15 14 18 15 5 Go 69 69 70 71 73 72 70 69 70 70 Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 36 Cu 70 60 70 10 10 20 10 10 10 Ga 7 6 5 2 2 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 166 Rb 29 6 4 3 2 2 3 2 2 Sr 10 12 9 2 <t< th=""><td>P₂O₅</td><td>0.03</td><td>0.03</td><td>0.03</td><td>0.01</td><td>0.01</td><td>0.03</td><td>0.02</td><td>0.03</td><td>0.03</td><td>0.03</td></t<> | P ₂ O ₅ | 0.03 | 0.03 | 0.03 | 0.01 | 0.01 | 0.03 | 0.02 | 0.03 | 0.03 | 0.03 |
| Ba 32 30 37 9 10 15 14 18 15 Co 69 69 70 71 73 72 70 69 70 70 Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 36 Cu 70 60 70 10 10 20 10 10 10 10 Ga 7 6 5 2 2 8 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 16 Rb 29 6 4 3 2 2 3 2 2 5r Sr 10 12 9 2 2 4 3 4 4 4 V 78 80 67 5 5 48 46 45 48 46 Y 34 25 33 2 2< | LOI | 4.42 | 4.65 | 4.21 | 4.81 | 4.79 | 5.22 | 5.24 | 5.33 | 5.27 | 5.1 |
| Ba 32 30 37 9 10 15 14 18 15 Co 69 69 70 71 73 72 70 69 70 70 Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 36 Cu 70 60 70 10 10 20 10 10 10 Ga 7 6 5 2 2 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 166 Rb 29 6 4 3 2 2 3 4 4 V 78 80 67 5 48 46 45 48 46 Y 34 25 33 2 2 10 10 11 11 < | Total | 100.96 | 99.7 | 100.59 | 101.57 | 100.56 | 100.69 | 100.66 | 99.8 | 100.41 | 99.42 |
| Ba 32 30 37 9 10 15 14 18 15 Co 69 69 70 71 73 72 70 69 70 70 Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 36 Cu 70 60 70 10 10 20 10 10 10 Ga 7 6 5 2 2 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 166 Rb 29 6 4 3 2 2 3 4 4 V 78 80 67 5 48 46 45 48 46 Y 34 25 33 2 2 10 10 11 11 < | | | | | | | | | | | |
| Co 69 69 70 71 73 72 70 69 70 70 Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 366 Cu 70 60 70 10 10 20 10 10 10 Ga 7 6 5 2 2 8 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 166 Rb 29 6 4 3 2 2 3 2 2 Sr 10 12 9 2 2 4 3 4 4 V 78 80 67 5 48 46 45 48 4 Y 34 25 33 2 10 10 11 11 1 < | Trace eler | ment (µg/g) |) | | | | | | | | |
| Cr 1880 1820 1870 1030 1190 3770 3720 3640 3670 36 Cu 70 60 70 10 10 20 110 10 10 110 10 10 11 10 10 11 11 10 10 11 11 10 10 11 11 11 10 10 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 </th <td>Ba</td> <td>32</td> <td>30</td> <td>37</td> <td>9</td> <td>10</td> <td>15</td> <td>14</td> <td>18</td> <td>15</td> <td>15</td> | Ba | 32 | 30 | 37 | 9 | 10 | 15 | 14 | 18 | 15 | 15 |
| Cu 70 60 70 10 10 20 10 10 10 10 Ga 7 6 5 2 2 8 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 166 Rb 29 6 4 3 2 2 3 2 2 Sr 10 12 9 2 2 4 3 4 4 V 78 80 67 5 48 46 45 48 Y 34 25 33 2 2 10 11 11 11 11 Zr 17 14 13 8 8 20 11 15 17 15 Zn 50 60 60 40 70 220 170 160 170 14 | Со | 69 | 69 | 70 | 71 | 73 | 72 | 70 | 69 | 70 | 69 |
| Ga 7 6 5 2 2 8 8 8 8 Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 1660 Rb 29 6 4 3 2 2 3 2 2 Sr 10 12 9 2 2 4 3 4 4 V 78 80 67 5 5 48 46 45 48 Y 34 25 33 2 2 10 10 11 11 11 Zr 17 14 13 8 8 20 11 15 17 160 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 K/Rb 0.002 0.001 0.002 0.001 0.002 0.001 0.003 | Cr | 1880 | 1820 | 1870 | 1030 | 1190 | 3770 | 3720 | 3640 | 3670 | 3610 |
| Ni 1110 1080 1150 1670 1750 1710 1650 1620 1650 1660 Rb 29 6 4 3 2 2 3 2 2 Sr 10 12 9 2 2 4 3 4 4 V 78 80 67 5 5 48 46 45 48 Y 34 25 33 2 2 10 11 11 11 Zr 17 14 13 8 8 20 11 15 17 Zn 50 60 60 40 70 220 170 160 170 14 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 Rb/Sr 2.9 0.5 0.44 1.5 1 0.5 1 0.5 | Cu | 70 | 60 | 70 | 10 | 10 | 20 | 10 | 10 | 10 | 10 |
| Rb 29 6 4 3 2 2 3 2 2 Sr 10 12 9 2 2 4 3 4 4 V 78 80 67 5 5 48 46 45 48 Y 34 25 33 2 2 10 10 11 | Ga | 7 | 6 | 5 | 2 | 2 | 8 | 8 | 8 | 8 | 8 |
| Sr 10 12 9 2 2 4 3 4 4 V 78 80 67 5 5 48 46 45 48 Y 34 25 33 2 2 10 10 11 11 Zr 17 14 13 8 8 20 11 15 17 Zn 50 60 60 40 70 220 170 160 170 14 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 K/Rb 0.002 0.001 0.002 0.001 0.002 0.001 0.003 0.003 0.033 0.025 0.00 K/Ba 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.004 0.003 0.00 K/Ba 0.02 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.72 | Ni | 1110 | 1080 | 1150 | 1670 | 1750 | 1710 | 1650 | 1620 | 1650 | 1640 |
| V 78 80 67 5 5 48 46 45 48 Y 34 25 33 2 2 10 10 11 11 Zr 17 14 13 8 8 20 11 15 17 Zn 50 60 60 40 70 220 170 160 170 14 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 K/Ba 0.002 0.001 0.002 0.002 0.002 0.001 0.003 0.00 Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.72 Rb/Y 0.85 0.24 0.12 1.5 1 | Rb | 29 | 6 | 4 | 3 | 2 | 2 | 3 | 2 | 2 | 2 |
| Y 34 25 33 2 2 10 10 11 11 Zr 17 14 13 8 8 20 11 15 17 Zn 50 60 60 40 70 220 170 160 170 14 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 Rb/Sr 2.9 0.5 0.44 1.5 1 0.5 1 0.5 0.5 0.0 K/Ba 0.002 0.001 0.002 0.001 0.002 0.001 0.004 0.003 0.00 Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.72 Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.14 Na | Sr | 10 | 12 | 9 | 2 | 2 | 4 | 3 | 4 | 4 | 4 |
| Zr 17 14 13 8 8 20 11 15 17 Zn 50 60 60 40 70 220 170 160 170 160 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 Rb/Sr 2.9 0.5 0.44 1.5 1 0.5 1 0.5 0.5 0.0 K/Ba 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.004 0.003 0.00 Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.727 Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.182 0.182 Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.84 | V | 78 | 80 | 67 | 5 | 5 | 48 | 46 | 45 | 48 | 45 |
| Zn 50 60 60 40 70 220 170 160 170 140 K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.00 Rb/Sr 2.9 0.5 0.44 1.5 1 0.5 1 0.5 0.5 0.0 K/Ba 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.003 0.00 Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.72 Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.182 0.14 Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.84 | Y | 34 | 25 | 33 | 2 | 2 | 10 | 10 | 11 | 11 | 11 |
| K/Rb 0.002 0.007 0.0021 0.003 0.008 0.013 0.003 0.033 0.025 0.007 Rb/Sr 2.9 0.5 0.44 1.5 1 0.5 1 0.5 0.5 0.0 K/Ba 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.003 0.00 Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.72 Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.182 0.182 Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.84 | Zr | 17 | 14 | 13 | 8 | 8 | 20 | 11 | 15 | 17 | 18 |
| Rb/Sr 2.9 0.5 0.44 1.5 1 0.5 1 0.5 0.5 0.6 K/Ba 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.004 0.003 0.001 Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.72 Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.182 0.182 Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.84 | Zn | 50 | 60 | 60 | 40 | 70 | 220 | 170 | 160 | 170 | 190 |
| K/Ba 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.002 0.001 0.004 0.003 0.003 Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.72 Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.182 0.182 Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.84 | K/Rb | 0.002 | 0.007 | 0.0021 | 0.003 | 0.008 | 0.013 | 0.003 | 0.033 | 0.025 | 0.004 |
| Ga/Y 0.2 0.24 0.152 1 1 0.8 0.8 0.727 0.727 0.727 Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.182 0.182 Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.89 | Rb/Sr | 2.9 | 0.5 | 0.44 | 1.5 | 1 | 0.5 | 1 | 0.5 | 0.5 | 0.5 |
| Rb/Y 0.85 0.24 0.12 1.5 1 0.2 0.3 0.182 0.182 0.182 Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.89 | K/Ba | 0.002 | 0.001 | 0.0002 | 0.001 | 0.002 | 0.002 | 0.001 | 0.004 | 0.003 | 0.001 |
| Na/K 0.6 0.178 9.83 2.675 1.337 2.38 6.25 0.111 1.042 0.84 | Ga/Y | 0.2 | 0.24 | 0.152 | 1 | 1 | 0.8 | 0.8 | 0.727 | 0.727 | 0.727 |
| | Rb/Y | 0.85 | 0.24 | 0.12 | 1.5 | 1 | 0.2 | 0.3 | 0.182 | 0.182 | 0.182 |
| | Na/K | 0.6 | 0.178 | 9.83 | 2.675 | 1.337 | 2.38 | 6.25 | 0.111 | 1.042 | 0.892 |
| Ba/Rb 1.1 5.0 9.25 3.0 5.0 7.5 4.66 9.0 7.5 7 | Ba/Rb | 1.1 | 5.0 | 9.25 | 3.0 | 5.0 | 7.5 | 4.66 | 9.0 | 7.5 | 7.5 |

 Table 3: Major (%) and trace element composition of talc schist of Wonu-Apomu and Ilesa areas

1–5: Wonu-Apomu samples 6–10: Ilesa samples

Baba-Ode and Erin-Omu talc contents are ca. 49 % and 59 % respectively. Chlorite, on the other hand, is 14–20 % for Ilesa and Wonu-Apomu, while those of Oke-Ila and Iseyin are (18–22 %) and (12–22 %) respectively. It must be noted that pyrophyllite and anthophyllite, which occur prominently in the Ilesa and Wonu-Apomu samples are absent in the Iseyin and Oke-Ila samples.

Anthophyllite, an orthorhombic amphibole is a major constituent in the Ilesa and the Apomu talc-schist while tremolite, a monoclinic amphibole, which is one of the major constituents in the Oke-Ila, Iseyin, Baba-Ode and Erin-Omu talc-schist are absent in the Wonu-Apomu and Ilesa samples (Table 2). Anthophyllite is stable only at low temperatures. When heated to about 400 °C, it may alter to a monoclinic amphibole, such as, tremolite. The heat for the conversion of the anthophyllite to tremolite in the areas where they occur could have been provided by the granitic intrusives around the talc bodies.

Geochemistry

Chemical data show that the talcose rocks of the study areas are highly siliceous with an average value of 55.62 % and 52.35 % SiO₂ for Wonu-Apomu and Ilesa samples, respectively (Table 3). Also magnesia contents ranges between 27.89-30.06 % and 26.02-26.67 % for the Wonu-Apomu and Ilesa talcose samples. Alumina (0.77-4.3 %), Fe₂O₂ (3.75-8.345 %) and CaO (0.05-3.17 %) contents of Wonu-Apomu talcose rock show lower values and a wider range within samples when compared to the Ilesa samples with 4.99-5.55 %, 9.26-9.54 % and 1.31-1.39 % for Al₂O₂, Fe₂O₂ and CaO contents, respectively. Some samples from Wonu-Apomu, however, exhibit high silica (> 61 %) and magnesia contents (> 30 %), but correspondingly low Al_2O_3 (< 2 %), $Fe_2O_{3(t)}$ (< 4 %), CaO (0.05 %) and MnO contents (Table 4). These samples are higher in talc content than others (Tables 2, 3 and 4). The abundances of SiO₂, MgO, Al_2O_3 , $Fe_2O_{3(t)}$ and CaO between Wonu-Apomu and Ilesa samples invariably reflect subtle mineralogical differences between samples. The MnO, Na₂O, K₂O, TiO_2 and P_2O_5 concentrations are generally low

(< 0.05 %) and are not strikingly varied within samples. The loss on ignition (LOI) does not generally exceed 5.4 %.

As observed from Tables 3 and 4, Ilesa talcose samples are enhanced in Cr (> 3 600 μ g/g), Ni (> 1 620 μ g/g) and Zn(> 160 μ g/g), while Wonu-Apomu indicated an average concentrations of 1 558 μ g/g, 1 353 μ g/g and 56 μ g/g for Cr, Ni and Zn, respectively. Other trace elements such as Ba, Co, Cu, Ga, Rb, Sr, V, Y and Zr are generally low and do not show any marked trend. The talcose rock showed SiO₂ enrichment, probably due to the effect of chemical weathering. Hydrothermal alteration accompanied by serpentinization of the tholeiitic protoliths was reflected in high MgO (26.34-29.15 %) and $Fe_2O_{3(t)}$ (6.43–9.39 %) contents of the talcose rocks. The relative chemical mobility of Na, Ca and K during secondary alteration processes is largely displayed by strong depletions of these Na, Ca and K oxides in the talcose samples relative to the protolith amphibolites (Table 5) ^[22]. The Ba (24 μ g/g), Sr (7 μ g/g) and Rb (9 μ g/g) concentrations in the talcose rocks are generally low compared to the amphibolites within the area (Table 5). This trend is also due to their chemical instability during secondary alteration processes. On the other hand, Cr $(3.684 \mu g/g)$ and Ni $(1.654 \mu g/g)$ contents, as a result of their chemical immobility even under hydrothermal alteration, which commonly produce talc, serpentine and chlorite from mafic and ultramafic rocks showed distinctive chemical enrichment trends in the talcose rocks. The values for Co (ca. 70 μ g/g) remain unchanged in both the talcose rock and the amphibolites (Table 4). These could serve as exploration guide for basemetals and platinum group elements (PGE) in the area.

Functional potentials of the talc bodies

The physical and industrial properties of the talcose body include determination of natural moisture content (NMC), loss on ignition (LOI), linear shrinkage (LSK), water absorption capacity (WAC), firing colour and pH. Results obtained from these tests (Table 6) served as basis for the evaluation of functional potentials of the talc bodies. The results showed that the NMC ranged from 5.00 % to 6.55 %.

| | Wonu-Apomu | | | Ilesa | Isey | in (TTCS) | SW Nig | eria (TTAS) |
|-----------------------------------|------------|-------------------|-----------|----------------|------------|-------------|------------|-------------|
| | Mean | Range | Mean | Range | Mean | Range | Mean | Range |
| SiO ₂ | 55.62 | 51.23-61.15 | 52.35 | 52.15-52.82 | 54.70 | 53.61-55.35 | 55.01 | 46.82-55.37 |
| Al ₂ O ₃ | 2.74 | 0.77-4.3 | 5.17 | 4.99-5.3 | 3.54 | 1.86-4.84 | 2.52 | 1.19-2.86 |
| Fe ₂ O _{3(t)} | 6.43 | 3.75-8.34 | 9.39 | 9.26-9.54 | 6.50 | 5.75-7.25 | 4.50 | 3.20-4.50 |
| MnO | 0.19 | 0.05-0.48 | 0.14 | 0.14-0.15 | 0.16 | 0.10-0.25 | 0.05 | 0.004-0.13 |
| MgO | 29.15 | 27.89-30.06 | 26.34 | 26.02-26.67 | 27.20 | 22.06-30.38 | 30.04 | 29.13-32.04 |
| Ca0 | 1.71 | 0.05-3.17 | 1.36 | 1.31-1.39 | 4.43 | 2.76-5.32 | 1.50 | 0.41-4.47 |
| Na ₂ 0 | 0.04 | 0.01-0.11 | 0.05 | 0.01-0.08 | 0.22 | 0.16-0.32 | 0.01 | 0.01-0.02 |
| K ₂ 0 | 0.03 | 0.01-0.11 | 0.04 | 0.01-0.08 | 0.03 | 0.01-0.06 | 0.02 | 0.01-0.02 |
| TiO ₂ | 0.16 | 0.02-0.2 | 0.10 | 0.09-0.11 | Nd | Nd | Nd | Nd |
| P ₂ O ₅ | 0.02 | 0.01-0.03 | 0.03 | 0.02-0.03 | 0.02 | 0.02-0.3 | 0.06 | 0.05-0.13 |
| LOI | 4.58 | 4.21-4.81 | 5.23 | 5.1-5.33 | 2.99 | 1.82-4.88 | 6.00 | 3.60-5.41 |
| Total | 100.68 | | 100.2 | | 99.86 | | 99.71 | |
| T | | (-) | | | | | | |
| | ment (µg/ | | 15 | 14 10 | NJ | NJ | NJ | NJ |
| Ba | 24 | 9-37 | 15 | 14-18 | Nd | Nd | Nd | Nd |
| Co | 1550 | 69-73 | 70 | 69-70 | 71 | 52-80 | Nd 2000 | Nd |
| Cr | 1558 44 | 1030-1880 | 3684 | 3610-3770 | 826 | 806-897 | | Nd |
| Cu Ga | 44 | 10-70 2-7 | 12 8 | 10-20 | Nd | Nd Nd | Nd Nd | Nd Nd |
| Ni | 1352 | | | 0 1620-1710 | Nd 1278 | 1034–1702 | 1500 | |
| Rb | 9 | 1080–1750 2–29 | 1654 2 | 2-3 | Nd | Nd | Nd | Nd Nd |
| Sr | 7 | 2-29 | 4 | 3-4 | Nd | Nd | Nd | Nd |
| V V | 47 | 5-80 | 4 | 45-48 | Nd | Nd | Nd | Nd |
| <u>v</u> Y | 19 | 2-34 | 40 | 45-46 | Nd | Nd | Nd | Nd |
| Zr | 19 | 8-17 | 11 | 11-20 | Nd | Nd | Nd | Nd |
| Zn | 56 | 40-70 | 182 | 160-220 | 69 | 58-82 | Nd | Nd |
| K/Rb | 0.004 | 0.001-0.008 | 0.0155 | | Nd | Nd | Nd | Nd |
| Rb/Sr | 1.27 | 0.44-2.9 | 0.0100 | 0.5-1.0 | Nd | Nd | Nd | Nd |
| K/Ba | 0.011 | 0.001-0.002 | 0.002 | 0.004-0.006 | Nd | Nd | Nd | Nd |
| Ga/Y | 0.52 | 0.15-1.0 | 0.756 | 0.727-0.8 | Nd | Nd | Nd | Nd |
| Rb/Y | 0.74 | 0.24-1.5 | 0.21 | 0.182-0.3 | Nd | Nd | Nd | Nd |
| Na/K | 2.92 | 0.18-9.83 | 2.136 | 0.111-6.25 | Nd | Nd | Nd | Nd |
| Ba/Rb | 4.67 | 1.10-9.25 | 7.23 | 4.66-9.00 | Nd | Nd | Nd | Nd |

Table 4: Average chemical composition of Wonu-Apomu and Ilesa talcose rocks compared with other talcose rocks elsewhere in Nigeria

Nd: Not determined TTCS: Talc-tremolite schist, Iseyin ^[14] TTAS: Talc-tremolite/actinolite schist, SW Nigeria ^[13]

| | | *Amph | ibolite | Talcose rocks | | | | |
|--|-------|-------------|---------|---------------|--------|-------------|--------|-------------|
| | Won | u-Apomu | Ilesa | | Woni | ı-Apomu | I | lesa |
| | Mean | Range | Mean | Range | Mean | Range | Mean | Range |
| SiO ₂ | 48.91 | 44.79-50.62 | 47.27 | 44.86-49.10 | 55.62 | 51.23-61.15 | 52.35 | 52.15-52.82 |
| Al ₂ O ₃ | 15.11 | 14.01-16.41 | 17.29 | 15.20-20.22 | 2.74 | 0.77-4.3 | 5.17 | 4.99-5.3 |
| Fe ₂ O _{3(t)} | 13.31 | 12.27-14.62 | 12.44 | 11.56-13.34 | 6.43 | 3.75-8.34 | 9.39 | 9.26-9.54 |
| MnO | 0.13 | 0.07-0.23 | 0.11 | 0.08-0.14 | 0.19 | 0.05-0.48 | 0.14 | 0.14-0.15 |
| MgO | 9.84 | 8.08-12.15 | 9.52 | 7.85-11.16 | 29.15 | 27.89-30.06 | 26.34 | 26.02-26.67 |
| CaO | 9.33 | 8.72-10.00 | 9.50 | 8.11-10.55 | 1.71 | 0.05-3.17 | 1.36 | 1.31-1.39 |
| Na ₂ O | 1.52 | 1.13-1.88 | 1.87 | 1.28-2.72 | 0.04 | 0.01-0.11 | 0.05 | 0.01-0.08 |
| K ₂ 0 | 0.73 | 0.46-0.99 | 0.732 | 0.55-0.85 | 0.03 | 0.01-0.11 | 0.04 | 0.01-0.08 |
| TiO ₂ | 0.05 | 0.25-0.88 | 0.61 | 0.39-0.86 | 0.16 | 0.02-0.2 | 0.10 | 0.09-0.11 |
| $P_{2}O_{5}$ | 0.03 | 0.01-0.05 | 0.03 | 0.02-0.05 | 0.02 | 0.01-0.03 | 0.03 | 0.02-0.03 |
| LOI | 0.31 | 0.25-0.35 | 1.46 | 0.24-0.35 | 4.58 | 4.21-4.81 | 5.23 | 5.1-5.33 |
| Total | 99.69 | | 99.86 | | 100.68 | | 100.20 | |

Table 5: Comparison of the major (%) and trace element (%) composition of the amphibolites and the talcose rocks in Wonu

 Apomu and Ilesa areas

Trace elements $(\mu g/g)$

| Ba | 184 | 86-350 | 137 | 88-231 | 24 | 9-37 | 15 | 14-18 |
|----|-----|---------|-----|---------|-------|-------------|-------|-------------|
| Sr | 123 | 106-143 | 122 | 106-145 | 7 | 2-12 | 4 | 3-4 |
| Rb | 33 | 26-40 | 34 | 27-46 | 9 | 2-29 | 2 | 2-3 |
| Cr | 79 | 58-92 | 63 | 45-94 | 1 558 | 1 030-1 880 | 3 684 | 3 610-3 770 |
| Со | 71 | 40-85 | 62 | 48-82 | 70 | 69-73 | 70 | 69–70 |
| Ni | 64 | 24-98 | 57 | 25-85 | 1 352 | 1 080-175 | 1 654 | 1 620-1 710 |
| Zr | 55 | 48-62 | 56 | 46-66 | 12 | 8-17 | 16 | 11-20 |
| Y | 37 | 26-45 | 37 | 25-45 | 19 | 2-34 | 11 | 10-11 |

* Wonu-Apomu and Ilesa amphibolites [29]

The LOI ranged between 4.21 % and 5.27 %. The LSK values are generally low (2.33–4.40 %). The WAC ranged from 8.05 % to 10.00 % while the pH of the slurry produced from the talcose rock ranged between 8.00 and 8.35 indicating alkalinity (Tables 6 and 7). The pH values and the LOI compare favourably with those of talc bodies from Erin-Omu, Iseyin, Oke-Ila and Baba-Ode (Table 7) all in southwestern Nigeria ^[15, 16, 18]. On the other hand the WAC is lower than those of the Erin-Omu and Baba-Ode samples and comparable to that of Oke-Ila but higher than that of Iseyin (Table 7). The variation in the mineralogical compositions and textures of the talcose rocks from one area to the other is reflected in the results of the firing tests presented. Generally, the results are comparable to other similar talc bodies within the schist belts of south-western Nigeria as reported by Durotoye and Ige^[11]. The colour of the raw talcose pellets changes from green, brown, grey or cream to shades of red, yellow and cream after firing (Table 6). Bleaching of the talcose rocks yielded white to off-white coloured samples with increasing concentration of the HCl acid used.

Based on the foregoing the talc bodies of Wonu-Apomu and Ilesha areas are assessed

as viable industrial raw materials in coloured ceramics including insulation ceramics^[23]. They could also function as filler and extender in paints ^[24], paper, roofing materials and textiles after screening to remove gritty particles. which are mainly quartz ^[25-27]. The talc bodies could be used to improve the stability and rigidity of plastic goods at high temperatures based on the specifications of Noble ^[28]. It must be noted that pyrophyllite is commonly used for the same purpose as talc. The high content of trace elements in the talc bodies, notably Cr. Ni, Co, Cu, V and Zn would not permit utilization for pharmaceutical purposes. Also, chemical data are not within the specifications of the American Society for Testing of Materials ^[29] for the manufacture of powder, creams and soaps. However, the bleaching test suggested that they could serve as absorbents for oil, grease and other chemicals.

Conclusions

Talcose rocks from Wonu-Apomu and Ilesa areas contain talc, pyrophyllite, anthophyllite and chlorite (clinochlore). Comparative studies of these talc bodies with those of Erin-Omu, Iseyin, Oke-Ila and Baba-Ode showed that the former is higher in pyrophyllite, anthophyllite and chlorite (clinochlore) but lower in talc and lack tremolite/actinolite contents.

Chemical data for the talcose rock showed that they are siliceous. Petrochemical trends of the major and trace element data of the talcose rocks reflected some subtle relationships to the amphibolites in the areas. Trace elements concentrations of the talcose rocks indicated enrichments in Cr, Ni, Cu, V and Zn relative to the amphibolites within the area. Petrogenetic indices suggest tholeiitic basalt precursor for the talc bodies ^[29].

| | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 |
|--------------|-------|-------|--------|-------|-------|-------|--------|-------|-------|-------|
| NMC | 5.20 | 5.35 | 5.20 | 6.20 | 6.55 | 5.00 | 5.26 | 5.10 | 5.44 | 5.00 |
| LOI | 4.42 | 4.65 | 4.21 | 4.81 | 4.79 | 5.22 | 5.24 | 5.33 | 5.27 | 5.10 |
| LSK | 4.40 | 3.83 | 4.27 | 2.40 | 2.33 | 2.54 | 3.60 | 3.38 | 3.42 | 3.33 |
| WAC | 8.54 | 8.33 | 8.05 | 9.48 | 10.00 | 8.05 | 8.40 | 8.55 | 8.40 | 8.35 |
| Raw colour | Brown | Brown | Green | Cream | Cream | Brown | Green | Brown | Brown | Brown |
| Fired colour | Red | Red | Yellow | White | White | Red | Yellow | Red | Red | Red |
| рН | 8.35 | 8.30 | 8.33 | 8.00 | 8.10 | 8.30 | 8.26 | 8.34 | 8.30 | 8.25 |

NMC – Natural moisture content (%)

LOI – Loss on ignition (%)

LSK – Linear shrinkage (%)

WAC – Water absorption capacity (%)

Table 7: Result of physical and industrial test on Wonu-Apomu and Ilesha talcose rocks compared to similar rock types in SW

 Nigeria

| | This study | Erin-Omu | Iseyin | Oke-Ila | Baba-Ode |
|-----|------------|-------------|------------|-----------|-------------|
| NMC | 5.00-6.55 | Nd | Nd | 7.50-8.90 | Nd |
| LOI | 4.21-5.27 | 2.87-4.11 | 1.45-6.03 | 3.75-5.28 | 4.25-4.41 |
| LSK | 2.33-4.40 | 1.01-1.81 | 0.25-2.0 | 2.45-3.50 | 1.01-1.52 |
| WAC | 8.05-10.00 | 13.25-16.25 | 6.96-11.65 | 4.21-5.80 | 15.14-18.25 |
| рН | 8.00-8.35 | 8.01-8.21 | Nd | 7.5-9.5 | 8.01-8.21 |

This study – Wonu-Apomu and Ilesa areas

NMC – Natural moisture content (%)

LOI – Loss on ignition (%) LSK – Linear shrinkage (%)

WAC – Water absorption capacity (%)

The Wonu-Apomu and Ilesa talc bodies possess requisite mineralogical, chemical and industrial specifications for use as raw materials in the manufacture of ceramics, paints, rubber, paper, textile, plastic, roofing materials and textiles. They can also be used as absorbent. However, they might not be appropriate for use in the pharmaceutical, powder, creams and soaps manufacturing industries without rigorous beneficiation, which might not be economical.

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